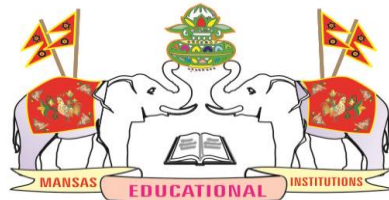# ACADEMIC REGULATIONS & CURRICULUM

**Applicable to the students admitted from the
Academic Year 2020-2021**

MECHANICAL ENGINEERING
(B.Tech. Programme)

MAHARAJ VIJAYARAM GAJAPATHI RAJ COLLEGE OF ENGINEERING
(Autonomous)

(Approved by AICTE, New Delhi, and permanently affiliated to JNTUK, Kakinada)

Listed u/s 2(f) & 12(B) of UGC Act 1956.

Vijayaram Nagar Campus, Chintalavalasa,Vizianagaram-535005, Andhra Pradesh

The visionaries

Late Dr. P V G Raju
Raja Saheb of Vizianagaram
Founder Chairman-MANSAS
Ex-Minister for Education and Health, Govt. of AP
Ex Member of Parliament

Late Dr. P. Anand Gajapathi Raju
Ex-Chairman-MANSAS
Ex-Minister for Education and Health
Govt. of AP
Ex Member of Parliament

P. Ashok Gajapathi Raju
Chairman-MANSAS
Ex-Union Minister for Civil Aviation,
Govt. of India
Ex-Minister for Finance, Govt. of AP

Vision

Maharaj Vijayaram Gajapathi Raj College of Engineering strives to become a centre par excellence for technical education where aspiring students can be transformed into skilled and well-rounded professionals with strong understanding of fundamentals, a flair for responsible innovation in engineering practical solutions applying the fundamentals, and confidence and poise to meet the challenges in their chosen professional spheres.

Mission

The management believes imparting quality education in an atmosphere that motivates learning as a social obligation which we owe to the students, their parents/guardians and society at large and hence the effort is to leave no stone unturned in providing the same with all sincerity. Towards that end, the management believes special focus has to be on the following areas:

M1: Have on-board staff with high quality experience and continuously updating themselves with latest research developments and sharing that knowledge with students.

M2: Having a well stream-lined teaching learning process that is continuously assessed for effectiveness and fine-tuned for improvement.

M3: Having state-of-the-art lab and general infrastructure that gives students the necessary tools and means to enhance their knowledge and understanding.

M4: Having a centralized department focused on improving placement opportunities for our students directly on campus and coordinating the training programs for students to complement the curriculum and enhance their career opportunities.

M5: Having advanced research facilities and more importantly atmosphere to encourage students to pursue self-learning on advanced topics and conduct research.

ABOUT THE INSTITUTION:

        Maharajah Alak Narayan Society of Arts and Science (MANSAS) is an Educational Trust founded by Dr. (late) P.V.G Raju, Raja Saheb of Vizianagaram in the hallowed memory of his father Maharajah Alak Narayan Gajapati with a view to confound socio-economic inequalities in the Vizianagaram principality executing a trust deed on 12-11-1958 duly established Maharajah's College and other educational institutions in and around Vizianagaram. The Trust is a charitable one published under Section 6 a (1) of A.P Charitable and Hindu Religious Institutions and Endowment Act 30 of 1987.

        The object of the Trust is to manage the properties of educational institutions under it and to promote and advance the cause of education in general, besides awarding scholarships to deserving students enabling them to undergo special training in science and industries in and out of India. The Trust has made an uncompromising contribution to the nation by presenting the stalwarts.

        Trust offers KG to PhD level education in Arts, Sciences, Law, Pharmacy, Humanities Education, Engineering and Management and presently houses 13 Educational Institutions. MVGR College of Engineering is one of the 13 Institutes.

Other Institutions under MANSAS

1. M.R. HIGH SCHOOL 1857
2. M.R COLLEGE (**NAAC ACCREDITED**) 1879
3. M.R. COLLEGE OF EDUCATION 1950
4. M.R. WOMENS COLLEGE (**NAAC ACCREDITED**) 1962
5. M.R. GIRLS HIGH SCHOOL 1974
6. M.R. MODEL HIGH SCHOOL 1974
7. M.R. ENGLISH MEDIUM SCHOOL 1979
8. M.R.V.R.G.R LAW COLLEGE 1987
9. M.R. P.G. COLLEGE (**NAAC ACCREDITED**) 1987
10. M.R.SCHOOL OF MANAGEMENT STUDIES 1994
11. M.R.V.R.G.R – II MEMORIAL JR. COLLEGE 1994
12. M.R. COLLEGE OF PHARMACY 2004

Maharaj Vijayaram Gajapathi Raj (MVGR) College of Engineering was established in the year 1997 by Maharaj Alak Narayan Society for Arts and Sciences (MANSAS) to impart quality technical education. The Institution is located in lush green, serene and pollution free environment spread over 60 acres of land in Chintalavalasa village situated in the outskirts of Vizianagaram, a fort city in the north coastal region of Andhra Pradesh.

Institution at a glance:
- MVGR is a 23 years old institution, established in 1997
- All eligible UG Programs (CHEMICAL, CIV, CSE, ECE, EEE, IT & MECHANICAL) were reaccredited by NBA.
- MBA program was also re-accredited by NBA.
- Had been re-accredited with Grade 'A' by NAAC of UGC
- Has Permanent affiliation with JN Technological University-Kakinada
- Listed under sections 2(f) & 12(b) of UGC act 1956.
- Approved by AICTE-New Delhi
- EIGHT departments are recognized as RESEARCH CENTERS by JNTU-K
- Granted Autonomy by UGC in 2015
- Campus of 60 acre
- Offering 7 UG and 5 M.Tech., and 1 MBA program
- About 250 faculty of which 84 Ph.D. Degree holders
- 83 Laboratories with an investment of about 13 Crores
- Total built up area of about 7 Lakh Sft
- About 42,000 volumes and Access to 8 international online journal packages like IEEE, SPRINGER, etc.
- 1420 Systems & 395 Mbps band width internet facility
- About Rs. 4 Crore worth of on-going R&D projects
- Actively involved in civil engineering consultancy work as Third Party Quality Auditor for Vizianagaram Municipality
- WIPRO Recognized technology learning center and MISSION 10X partner institution
- Recognized National Instruments Academy for Training in LabView
- SIRO Recognition by DSIR
- Recognized PTC Centre of Excellence for Creo Training
- Identified by MSME as Business Incubation Centre
- APSSDC-Siemens Technical Skill Development Institute
- Recognized CMs SKILL EXCELLENCY CENTER (SEC)
- Microsoft Ed-vantage Platinum Partner
- Institutional member of IUCEE
- Institutional Member of CII
- Member, Chamber of Commerce, Vizianagaram
- Green Campus award by Govt. of AP

MVGR College of Engineering is rated as one among the best engineering colleges in the state of Andhra Pradesh as it set up highest standards in all areas of curricular, co-curricular and extra-curricular activities and in students' placements. Based on industry and expert's feedback, the college is updating the curriculum from time to time. The college offers many value added add-on courses students and conducts training programs to meet the industries' requirements.

Academic Regulations for B.Tech., Program

Applicable to the students admitted from the Academic year 2020-2021 onwards.

1. PROGRAM STRUCTURE:

B.Tech.:

Sl. No	Category	Credits
1	Humanities and Social Sciences including Management courses	12
2	Basic Science courses	25
3	Engineering Science courses including workshop, drawing, basics of electrical/mechanical/computer etc.	26
4	Professional core courses	54
5	Professional Elective courses relevant to chosen specialization/branch	18
6	Open subjects – Electives from other technical and /or emerging subjects	12
7	Project work, seminar and internship in industry or elsewhere	13
8	Mandatory Courses [Environmental Sciences, Induction training, Indian Constitution, Essence of Indian Traditional Knowledge]	0
	Total	160

- Open electives offered by the parent department are listed in the course structure and are offered to students of other programs. The students of parent departments may also opt the course, provided it shall not be listed in the curriculum.

- For audit course a student is deemed to satisfy the minimum contact hours, as prescribed by the department and shall also comply with the requirements for submission of assignments/projects. A student shall also opt for MOOCs and submit the certificate.

1. HSS Courses		
Sl. No.	Subject	Credits
1	English -1	3
2	English -2 (Technical English)	3
3	Elective-1 (Management Related course (MEFA or MS or Operations Research)	3
4	Elective-2 (Professional Ethics and Human Values)	3
	Total	**12**

2. Basic Science Courses		
Sl. No.	Subject	Credits
1	Mathematics-I	3
2	Mathematics-II	3
3	Mathematics-III	3
4	Mathematics-IV	3
5	Applied / Engineering Physics	3
6	Applied / Engineering Physics Lab	2
7	Engineering Chemistry	3
8	Engineering Chemistry Lab	2
9	Biology for Engineers	3
	Total	**25**

3. Engineering Science Courses		
Sl. No.	Subject	Credits
1	Programming for Problem Solving	3
2	Programming for Problem Solving Lab	2
3	Internet of Things (IOT)	3
4	Computer aided Engineering Graphics	3
5	Basic Electrical Engineering	3
6	Basic Electrical Engineering Lab	2
7	Department wise Engineering Science Course-I AI Tools , Techniques & Applications	3
8	AI Tools , Techniques & Applications Lab	2
9	Department wise Engineering Science Course-II (Design thinking and Product Innovation)	3
10	Workshop (Department Specific)	2
	Total	**26**

	Subjects	Credits
1	Professional Core Courses	54
2	Professional Elective Courses Relevant to Chosen Specialization/Branch	18

3	Open Subjects – Electives from other Technical and / or Emerging Subjects	12
		84

7. Project		
Sl. No.	Subject	Credits
1	Socially Relevant Project	1
2	Mini Project	2
3	Project Phase - I	2
4	Project Phase - II	8
	Total	**13**

8. Audit Courses (Non Credit Course)		
Sl. No.	Subject	
1	Induction Program	
2	Constitution of India	
3	Essence of Indian Traditional Knowledge	
4	Environmental Science	

BOS Chairman shall notify the list of MOOCs offered (Open Elective & Professional Elective) in the beginning of the semester.

2. PROGRAM PATTERN:

B.Tech.: The program is for 4 academic years / 8 semesters.

B.Tech. (Lateral Entry): The program is for 3 academic years / 6 semesters.

3. AWARD OF DEGREE:

B.TECH:

A student will be declared eligible for the award of degree if he/she fulfills the following academic regulations.

a) A student shall be declared eligible for the award of degree, if he/she pursues a course of study for not less than four academic years and not more than eight academic years from the date of admission.

b) The student shall register for **160** credits and secure all **160** credits.

c) The medium of instruction for the entire under graduate program in Engineering & Technology will be in **English** only.

d) Skill oriented courses are embedded with domain specific, communication and Advanced / elective courses.

e) A student shall also register and successfully complete audit programs (Non-credit) as recommended by Academic Council.

f) A student on completing 1st year class work may opt for a break of 1 year which shall

be deemed as GAP year, as recommended by APSCHE, for undertaking successful entrepreneurial ventures.

g) Students who fail to complete Four Years Course of study within 8 years shall forfeit their seat and their admission shall stand cancelled.

B.TECH. (Honors)/(Minors) :

B.Tech. with Honors or B.Tech. with Minor will be awarded if the student earns 20 additional credits as per the regulations/guidelines.

B.TECH (Lateral Entry):

A student will be declared eligible for the award of degree on fulfilling the following academic requirements.

a) A student shall be declared eligible for the award of the degree, if he/she pursues a course of study for not less than three academic years and not more than six academic years.

b) The student shall register for **126** credits and secure all **126** credits.

c) A student shall also register and successfully complete audit programs (Non-credit) as recommended by Academic Council.

d) Students who fail to complete their three Years Course of study within 6 years shall forfeit their seat and their admission shall stand cancelled.

e) Student shall register for bridge programs, if any, as administered by the respective departments at the beginning of 2nd year and successfully complete as per the guidelines of the Institution.

B.TECH. (Honors)/(Minors) :

B.Tech. with Honors or B.Tech. with Minor will be awarded if the student earns 20 additional credits as per the regulations/guidelines.

4. CERTIFICATION PROGRAMS:

Sl. No.	Dept.	Name of the Program
1	MECH	Windchill 10.2 PDM by Adroitec Engineering Solutions Pvt. Ltd., Hyderabad
2	MECH	Creo 2.0 by PTC
3	MECH	Edgecam by Verosoft, UK
4	MECH	ANSYS Training and Certification by Mechanical Department
5	MECH	AUTOCAD Training and Certification by Mechanical Department
6	MECH	Catia by APSSDC-Dassault Systems, CM's Center of Excellence
7	MECH	Delmia by APSSDC-Dassault Systems, CM's Center of Excellence
8	MECH	Simulia by APSSDC-Dassault Systems, CM's Center of Excellence
9	MECH	2-Wheeler Automobile Certification by APSSDC-SIEMENS
10	MECH	4-Wheeler Automobile Certification by APSSDC-SIEMENS

11	MECH	Welding Certification by APSSDC-SIEMENS
12	MECH	CNC Certification by APSSDC-SIEMENS
13	MECH	Commercial Electrical Certification by APSSDC-SIEMENS
14	MECH	Solid Edge Certification by APSSDC-SIEMENS
15	CHEM	Chemical Process Design and Simulation by Simtech Simulations, Hyderabad
16	ECE	Embedded Systems by Think LABS, Mumbai
17	ECE	Labview by National Instruments Systems India Pvt. Ltd.
18	ECE	Unified Technology Learning Program (UTLP) by Wipro Mission 10X
19	CSE, IT	PEGA by Virtusa Corporation
20	CSE, IT	Microsoft technologies by Microsoft Corp.
21	CSE, IT	Ethical Hacking by EC-Council Academia
22	CSE, IT	Java and C by Talent Sprint
23	CSE, IT	Network Analyst (CCNA) by Cisco Systems Inc
24	CSE, IT	Java Programming (OCJP) and DBMS by Oracle
25	EEE	PLC, Drives and Automation by Siemens
26	EEE	PLC by New Dawn Automation
27	EEE	Home Electrical Certification by APSSDC-SIEMENS
28	Civil	Remote Sensing and GIS by Indian Institute of Remote Sensing

a) The Institution shall offer the certification programs by itself or in collaboration with industry/such other Institutions deemed to have specialized expertise in the proposed area of training.

b) Only students of the Institution shall be eligible to register on payment of prescribed fee.

c) However, subject to availability of resources and the demand the Institution may offer the program to external candidates meeting the pre-qualification requirements and in the order of the merit.

d) The duration of the course and design of the content shall be done by the respective departments of the Institution by themselves or in collaboration with industry/such other institutions deemed to have specialized expertise in the proposed area of training.

e) If the duration of the course is less than or equal to 40 hours, it can be completed in one semester, otherwise, it can suitably distributed over a number of semesters.

f) Mere enrolment/registration for the program shall not entitle any claim for award of certificate.

g) A candidate shall be deemed eligible for the award of the certificate if he/she
   - Attends at least 75% of scheduled training sessions
   - Complies to all the requirements of submission of the assignments, presentations, seminars, projects, etc., and also appears for periodic tests.
   - Shall attain minimum levels of performance in tests as prescribed.
   - Shall remit such fee as deemed fit for the certification
   - A candidate registered and failed to meet the requirements shall be permitted to repeat the said training one another time after remitting 25% of the fee fixed for the program as re-registration fee.

If the student is absent for the periodic tests, the test shall be re-conducted on payment of

10% of fee.

5. COURSES OFFERED:

Name of the Program	Degree
UG Programs (Engineering & Technology)	B.Tech. (Civil)
	B.Tech. (EEE)
	B.Tech. (Mech.)
	B.Tech. (ECE)
	B.Tech. (CSE)
	B.Tech. (CHEM)
	B.Tech. (IT)
PG Programs (Engineering & Technology)	M.Tech. (Structural Engineering)
	M.Tech. (Power Systems)
	M.Tech. (PDM)
	M.Tech. (VLSI)
	M.Tech. (CN&IS)
Other PG Programs	MBA
Research Programs	Ph.D. in Civil, EEE, MECH, ECE, CSE, CHEM, MBA and MATHS

6. DISTRIBUTION AND WEIGHTAGE OF MARKS:

B.Tech.:

a). **Theory :**

All Theory courses will have 5 units and assessed for 100 marks, of which, 30 marks for internal assessment and 70 marks for semester end external examination.

Internal Assessment:

Subjective tests   - 15 Marks
Objective tests   - 10 Marks
Assignments   - 05 Marks

- Two Mid – Term examinations (Cycle 1 and Cycle2) shall be conducted. One on first 50% of the syllabus and second on remaining 50% of the syllabus.

- Each Mid Term examination consists of a) Subjective test b) Objective test c) Assignment

- Each subjective test shall be conducted for 90 Minutes and have 3 questions each for 5 marks (No choice) and assessed for 15 marks

- Each objective test shall be conducted for 20 minutes and have 20 Multiple

Choice Questions each for 1/2 mark and assessed for 10 Marks.

- Assignments shall be assessed for 5 marks.

- The total marks assessed for each Mid – Term examination (Cycle 1 or Cycle 2) is 30 marks.

- Final internal marks can be calculated with 80% weightage for the better of the two Mid-Term examinations and 20% weightage for other Mid-Term examination.

Example:

Mid-Term 1 (Cycle 1) Marks = Marks secured in (Subjective test-1 + Objective test-1 + Assignment-1)

Mid-Term 2 (Cycle 2) Marks = Marks secured in (Subjective test-2 + Objective test-2 + Assignment-2)

Final internal marks = [Best of (Mid term-1/Mid - Term-2) marks X 0.8 + Least of (Mid Term-1/Mid Term-2) marks X 0.2]

External Assessment:

- External examination is for 70 marks (180 min). Question paper contains 10 questions (2 questions from each unit) and each question carries 14 marks. Student shall answer 5 questions (1 question from each unit).

i) Design Thinking and Product Innovation - Evaluation pattern

Internal Assessment: **30 Marks**

Project based learning	- 20 Marks
Assignments	- 10 Marks

Project based learning: The student has to identify a problem and provide a solution by applying design thinking methodologies and submit a report, which is assessed for 20 Marks.

Assignments: The student has to submit 4 assignments (1 for each unit) and assessed for 10 marks. Each assignment shall consist of 4 questions (4X10 = 40 marks) and the same shall be scaled down to 10 marks. Average of 4 assignments shall be considered as final assignment marks.

External Assessment: **70 Marks**

External examination is for 70 marks (180 min). Question paper contains 8 questions from first IV units (2 questions from each unit) and each question carries 12 marks. Student shall answer 4 questions from first IV units (1 question from each unit) and case study (22 Marks) from V unit.

ii) Internet of Things (IoT) , Surveying and Geomatics, MAT Lab Programming, Programming with Lab View, Embedded Processor - Evaluation pattern

Internal Assessment: **30 Marks**

Subjective Test	- 15 Marks
Project based learning	- 15 Marks

- Two subjective tests shall be conducted.
- Each subjective test shall be conducted for 90 Minutes and have 3 questions each for 5 marks (No choice) and assessed for 15 marks
- Project based learning shall be assessed for 15 Marks.

- In Project based learning, a student has to identify a problem such that at least 3 or 4 modular learning of experiments shall be integrated and submit comprehensive report with solution at the end of the semester.

- Final internal marks (30 Marks) can be calculated with 80% weightage for the better of the two subjective tests (15 Marks) and 20% weightage for other subjective test plus + Project based learning (15 Marks)
  .

External Assessment: 70 Marks

External examination is for 70 marks (180 min). Question paper contains 10 questions (2 questions from each unit) and each question carries 14 marks. Student shall answer 5 questions (1 question from each unit).

b). **Laboratory/Practice:**

All Laboratory/Practice courses are assessed for 50 marks, of which, 15 marks for internal assessment and 35 marks for semester end external examination.

Internal Assessment : (15 Marks)

Continuous assessment : 05 Marks
Project based learning   : 05 Marks
Internal test   : 05 Marks

- Continuous assessment for 05 marks for each experimental session finally averaged to 05 marks.
- Project based learning shall be assessed for 05 Marks.
- In Project based learning, a student has to identify a problem such that at least 3 or 4 modular learning of experiments shall be integrated and submit comprehensive report with solution at the end of the semester.
- An internal assessment test conducted at the end of the semester shall be assessed for 05 marks.

Semester End Assessment:

- Semester end examination is for 35 marks (180 min) conducted and assessed by both external and internal examiners.
- Both internal and external examination shall include assessment of the student on
  a) Knowledge of principles/concepts involved
  b) Experimental design
  c) Result interpretation and analysis

d) Experimental report

c). **Drawing/Design/Estimation:**

i) Computer Aided Engineering Graphics:

Evaluation Procedure:

The course will have 5 units and assessed for 100 marks, of which, 30 marks for internal assessment and 70 marks for semester end external examination.

Internal Assessment : (30 Marks)

Continuous assessment : 10 Marks
Project based learning   : 10 Marks
Internal Test   : 10 Marks

Semester End Assessment:

- Semester end examination is for 70 marks (180 min) conducted and assessed by both external and internal examiners.

- Question paper contains 3 questions (with internal choice). Each question carries 20 marks (5 marks for free hand drawing and list of commands & 15 marks for final drawing prepared in AUTOCAD). A Student shall answer all questions.

- Viva Voce for 10 Marks.

ii) Computer Aided Geometric Design and Assembly:

Evaluation Procedure:

The course will have 5 units and assessed for 50 marks, of which, 15 marks for internal assessment and 35 marks for semester end external examination.

Internal Assessment : (15 Marks)

Continuous assessment : 10 Marks
Project based learning   : 05 Marks

Semester End Assessment:

- Semester end examination is for 35 marks (180 min) conducted and assessed by both external and internal examiners.

- Semester End Examination shall include assessment of the student on Final drawings like modeling, assembly and drafting.

- Student is expected to execute one exercise.

- Final drawings like modeling, assembly and drafting hard copies shall be evaluated by both internal and external examiners

iii) Design and Drawing Courses

Evaluation Procedure:

The course will have 5 units and assessed for 100 marks, of which, 40 marks for internal assessment and 60 marks for semester end external examination.

Internal Assessment:	40 Marks
Subjective Test	- 20 Marks
Assignments	- 10 Marks
Design and Drawing reports	- 10 Marks

- Two subjective tests shall be conducted.
- Each subjective test shall be conducted for 90 Minutes and have 3 questions each for 7 marks (No choice) and the same shall be scaled down to 20 Marks.
- Average of two subjective tests shall be considered.
- Assignments shall be assessed for 10 marks.
- Design and drawing reports shall be assessed for 10 marks.

External Assessment:

The end examination question paper consists of Part A and Part B.

Part A consists of two questions regarding Design and Drawing (from two clusters clearly mentioned in the syllabus). Each question carries 20 marks. The student shall answer any 1 question.

Part B consists of four questions (from the remaining four clusters) with internal choice and all four are to be answered. Each question carries 10 marks.

iv) Estimation and Costing Courses

The course will have 5 units and assessed for 100 marks, of which, 40 marks for internal assessment and 60 marks for semester end external examination.

Internal Assessment:	40 Marks
Subjective Test	- 20 Marks
Assignments	- 10 Marks
Bar bending schedules,	- 10 Marks

Estimation and cost analysis reports

Two subjective tests shall be conducted.

- Each subjective test shall be conducted for 90 Minutes and have 3 questions each for 7 marks (No choice) and the same shall be scaled down to 20 Marks.
- Average of two subjective tests shall be considered.
- Assignments shall be assessed for 10 marks.
- Bar bending schedules, Estimation and cost analysis reports shall be assessed for 10 marks.

External Assessment:

External examination is for 60 Marks. The question paper consists of 2 questions. Each question carries 60 Marks. The student shall answer 1 question.

In each question, the section, plan and reinforcement drawings of various members of a building will be given and the following items are to be calculated.

- Quantities of all the items (20 Marks).
- Reinforcement tonnage and Bar bending schedule (10 Marks).
- Specifications (10 Marks).
- Rates of all the items as per Standard Schedule of Rates (20 Marks).

Non Credit Mandatory Courses: The student has to secure 40% of the marks allotted in the internal evaluation for passing the course (Satisfactory or Not-Satisfactory). No marks or letter grade shall be allotted.

Procedure for conduct and Evaluation of MOOCs.:
The student shall register for the (Minimum of 12 weeks) offered by SWAYAM/NPTEL as Program elective/Open elective with the approval of the Head of the Department. The Head of the Department shall appoint one mentor for each MOOC. The student has to register the course in SWAYAM/NPTEL portal. The student has to submit the pass certificate issued by SWAYAM/NPTEL after completion of the course.

Socially Relevant Project:

- A student shall identify and provide a solution to the problem relevant to society/Profession/Industry.

- A student shall engage at least 15 hours on socially relevant project. Socially relevant project shall be evaluated internally for 50 marks by Project Review Committee (PRC). PRC comprising of HoD, department Academic Coordinator, R&D member of the department, one senior faculty and guide shall review the progress.

Mini Project:

o A student shall undergo internship for a period of 4 weeks/provide solution to the problem relevant to Industry/ Modern tool during the vacation after VI semester and submit comprehensive report.

- Mini project shall be evaluated internally for 50 marks by Project Review Committee (PRC).

- PRC shall prepare rubrics for assessment.

Project Evaluation:

Project is divided into 2 phases – Phase I & Phase II and assessed for 200 Marks.

- Evaluation shall comprise of internal and external assessment.
  - Internal: 60 Marks
  - External: 140 Marks
- A project Review committee (PRC) comprising of HoD, department Academic Coordinator, R&D member of the department, one senior faculty and guide shall review the progress once in four weeks.

Project Phase I:

- Project Phase I shall be evaluated along with Project Phase II.
- A student shall undertake project phase I during the VII semester.
- A student shall report to the guide/external supervisor and work under his supervision at least 2 hours per week.
- Assessment shall be on
  - Literature review
  - Identification and statement of the Problem

Project Phase II:

- A student shall undertake project phase II during the VIII semester.
- A student shall report to the guide/external supervisor and work under his supervision at least 8 hours per week.
- Internal evaluation shall be done by HoD, department Academic Coordinator, R&D member of the department, one senior faculty and guide for 60 marks.
- External evaluation shall be done by HoD, Guide/Internal Examiner and External Examiner for 140 marks.
- Assessment shall be on
  a) Review on fundamental knowledge involved
  b) Inter disciplinary aspect
  c) Experimental/methodology design
  d) Result analysis and interpretations
  e) Report writing
  f) Team work
  g) Presentation
  h) Viva-voce

B.Tech. (Lateral Entry):

The rules and regulations for candidates admitted under lateral entry category for 2nd, 3rd and 4th years of study shall be same as applicable to regular B.Tech students.

7. ATTENDANCE REGULATIONS:

B.Tech.:

I. A student shall be eligible to appear for end semester examinations, if he or she acquires a minimum of 75% of attendance in aggregate of all the subjects (Theory & Lab.) for the semester.

II. Condonation of shortage of attendance in aggregate up to 10% (65% and above and below 75%) in each semester may be granted by the college academic committee.

III. Shortage of attendance below 65% in aggregate of all the subjects (Theory & Lab) for the semester shall not be Condoned.

IV. Detained student shall seek re- admission for that semester when offered within 4 weeks from the date of commencement of class work.

PROMOTION RULE (Based on attendance):

- A Student shall be promoted to the next semester on fulfillment of minimum attendance requirement (75%) of current semester.

PROMOTION RULE (Based on credits):

- A student shall be promoted from IV semester to V semester if he fulfills the minimum attendance requirement (75%) and academic requirement of 40% of credits up to either III semester or IV semester from the following examinations irrespective of whether the candidate takes the examination or not.
  - ➢ Two regular and Two supplementary examinations of I semester
  - ➢ Two regular and One supplementary examinations of II semester
  - ➢ One regular examination and One supplementary examination of III semester
  - ➢ One regular examination of IV semester.
- A student shall be promoted from VI semester to VII semester if he fulfills the minimum attendance requirement (75%) and academic requirement of 40% of credits up to either V Semester or VI semester from the following examinations irrespective of whether the candidate takes the examination or not.

- ➢ Three regular and Three supplementary examinations of I semester
- ➢ Three regular and Two supplementary examinations of II semester
- ➢ Two regular and Two supplementary examinations of III semester
- ➢ Two regular and One supplementary examinations of IV semester
- ➢ One regular and One supplementary examination of V semester
- ➢ One regular examination of VI semester.

B.TECH (Lateral Entry):

PROMOTION RULE (Based on attendance):

A Student shall be promoted to the next semester on fulfillment of minimum attendance requirement of current semester.

PROMOTION RULE (Based on credits):

A student shall be promoted from VI semester to VII semester if he fulfills the minimum attendance requirement (75%) and academic requirement of 40% of credits up to either V semester or VI semester from the following examinations irrespective of whether the candidate takes the examination or not.

- ➢ Two regular and Two supplementary examinations of III semester
- ➢ Two regular and one supplementary examinations of IV semester
- ➢ One regular and One supplementary examinations of V semester
- ➢ One regular examination of VI semester.

B.Tech. (Lateral Entry):

The rules and regulations for candidates admitted under lateral entry category for 2nd, 3rd and 4th years of study shall be same as applicable to regular B.Tech students.

8. MINIMUM ACADEMIC REQUIREMENTS:

B.Tech.: (Theory/Lab)

i. A student is deemed to have satisfied the minimum academic requirements for a course on securing minimum 35% of marks in the semester end exam and minimum 40% of marks in the sum total of the internal marks and semester end marks.

B.Tech. (Lateral Entry):

The rules and regulations for candidates admitted under lateral entry category for 2nd, 3rd and 4th years of study shall be same as applicable to regular B.Tech students.

9. GRADING SYSTEM:

B.Tech. / B.Tech. (Lateral Entry)

Semester Grade Point Average (SGPA) for the current semester which is calculated on the basis of grade points obtained in all courses, except audit courses and courses in which satisfactory or course continuation has been awarded,

SGPA= Σ (course credits earned x Grade points) /
Σ (Total course credits in the semester.

CGPA= Σ (course credits earned x Grade points) up to successfully completed semesters)   / Σ (Total course credits up to successfully completed)

The UGC recommends a 10-point grading system with the following letter grades as given below:

O	(Outstanding)	10
A+	(Excellent)	9
A	(Very Good)	8
B+	(Good)	7
B	(Above Average)	6
C	(Average)	5
P	(Pass)	4
F	(Fail)	0
Ab	(Absent)	0

- iii. A student obtaining Grade F shall be considered failed and will be required to reappear in the examination.

Marks Range Theory (Max – 100)	Marks Range Lab (Max – 50)	Level	Letter Grade	Grade Point
≥90	≥45	Outstanding	A+	10
≥80 to <89	≥40 to <44	Excellent	A	9
≥70 to <79	≥35 to <39	Very Good	B	8
≥60 to <69	≥30 to <34	Good	C	7

≥50 to <59	≥25 to <29	Fair	D	6
≥40 to <49	≥20 to <24	Satisfactory	E	5
<40	<20	Fail	F	0
-		Absent	AB	0

-

Illustration of Computation of SGPA and CGPA and Format for Transcripts

Computation of SGPA and CGPA

Illustration for SGPA

Course	Credit	Grade Letter	Grade point	Credit Point (Credit x Grade)
Course 1	3	A	8	3 X 8 = 24
Course 2	4	B+	7	4 X 7 = 28
Course 3	3	B	6	3 X 6 = 18
Course 4	3	O	10	3 X10 = 30
Course 5	3	C	5	3 X 5 = 15
Course 6	4	B	6	4 X 6 = 24
	20			**139**

Thus, **SGPA** =139/20 = 6.95

Illustration for CGPA

Semester 1	**Semester 2**	**Semester 3**	**Semester 4**	**Semester 5**	**Semester 6**
Credits: 16	Credits: 18	Credits: 25	Credits: 21	Credits: 23	Credits: 22
SGPA: 7.9	SGPA: 7.8	SGPA: 7.6	SGPA: 8.0	SGPA: 8.3	SGPA: 8.6

Semester 7	**Semester 8**
Credits: 21	Credits:14
SGPA: 8.2	SGPA: 8.5

Thus,

CGPA= $\frac{16\text{x}7.9 + 18\text{x}7.8 + 25\text{x}7.6 + 21\text{x}8.0 + 23\text{x}8.3 + 22\text{x}8.6 + 21\text{x}8.2 + 14\text{x}8.5}{160}$ = **8.1**

10. ELIGIBILITY FOR AWARD OF DEGREE:

B.Tech:

A student shall be eligible for award of the degree if he/she fulfills the following conditions:

1) Successfully completes all the courses prescribed for the Program.

2) CGPA greater than or equal to 5.0 (Minimum requirement for Pass),

11. AWARD OF CLASS:

B.Tech:

Eligible Candidates for the award of B.Tech., Degree shall be placed in one of the following Classes based on CGPA.

CLASS	CGPA	Remarks
First Class with Distinction	≥ 7.75 (Without any Supplemenary appearance)	From the CGPA secured from 126 credits from III semester to VIII semester
First Class	≥ 6.75	
Second Class	≥ 5.75 to < 6.75	
Pass Class	≥ 5.00 to < 5.75	

12. CURRICULAR FRAMEWORK FOR HONORS PROGRAMME:

The goal of introducing B.Tech (Hons) is to facilitate the students to choose additionally the specialized courses of their choice and build their competence in a specialized area in the UG level. The programme is a best choice for academically excellent students having good academic record and interest towards higher studies and research. All the students pursuing regular B.Tech with prerequisite CGPA are eligible to register for Honors degree course. A student has to acquire 20 more credits, in addition to 160 credits required, for the award of the B.Tech Honors degree ie 180 credits.. The additional courses shall be advanced subjects in the concerned department/discipline. The department concerned will determine required courses for award of Honor degree. The subjects in the Honor degree would be a combination of core (theory and lab) and some electives

OBJECTIVES:

The objectives of initiating the B. Tech (Honors) degree are:

> ➢ To encourage the undergraduate students towards higher studies and Research

> ➢ To prepare the students to acquire specialized knowledge in core Engineering streams

> ➢ To attain the high-level competence in the specialized area of Under Graduate program.

> ➢ To learn the best educational and professional skills in the specialized area.

> ➢ To provide the opportunity to learn the post graduate level courses in the specified undergraduate program.

Applicability and Enrolment:

> ➢ To all B. Tech (Regular and Lateral Entry) students admitted in Engineering & Technology

> ➢ The Honors degree will be offered at all JNTUK affiliated colleges (Autonomous and Non-Autonomous).

> ➢ The department offering Honors shall have at least one M. Tech in concerned stream, for B.Tech (Honors) registration.

> ➢ Total number of seats offered for a Honor program shall be a maximum of 35% of sanctioned intake .

> ➢ The allotment of seat into Honors degree is based on the percentage of marks obtained up to III semester in case of regular entry students and only III semester in case of lateral entry students

> ➢ In the event of any tie during the seat allotment for a Honors degree, the concerned major degree department offering Honor shall conduct a test/interview on the prerequisite subjects of Honors degree and final decision shall be taken.

> ➢ For applicability of Honors degree, both regular B Tech and Honors degree courses shall be successfully completed with specified SGPA/GCPA

> ➢ Transfer of credits from a particular minor to regular B. Tech or another major degree and vice versa shall not be permitted

> ➢ Institutions having at least two NBA accredited B.Tech/M.Tech programs can offer B.Tech(Honors) The Program departments should have valid NBA accreditation at the time of registration of the student for B.Tech (Honors).

Entry level:

> ➢ The B. Tech students (both Regular and Lateral Entry) pursuing a major degree

program can register for Honors degree.

- Students registering for Honors degree shall select the subjects from same branches/department based on the recommendations of BOS committee. For example, if a student pursumg major degree in Electrical & Electronics Engineering shall the selects subjects in Electrical & Electronics Engineering only and he/she will get major and Honors degree in Electrical & Electronics Engineering

- Only those students, who have a CGPA of 8.0 or above, without any backlog, will be permitted to register for a Honors degree

- An SGPA or CGPA in excess of 8.0 has to be maintained in the subsequent semesters in major as well as Honors degree without any backlogs in order to keep Honors degree registration active.

- Should both the SGPA and CGPA fall below 8.0 at any point after registering for the Honors, the Honors degree registration will cease to be active.

- A student registered for Honors degree in a discipline must register and pass in all subjects with a minimum GPA of 8.0 that constitute requirement for award of Honors degree.

- Separate SGPA/CGPA shall be shown on semester and final transcripts of regular B.Tech and Honor.

- Students shall not be permitted to register for Honors degree after completion of VI semester.

- Students shall be permitted to select a maximum of two subjects per semester from the list of subjects specified for Honors degree.

- The students shall complete Honors degree without supplementary appearance within stipulated period as notified by JNTUK for the completion of regular major B. Tech. program.

- Honors degree shall not be awarded at any circumstances without completing the regular major B Tech program in which a student got admitted

- If a student is detained due to lack of attendance, he/she shall not be permitted to register the courses for Honors degree

- If a student failed in any registered course of the Honors, he/she shall not be eligible to continue the B. Tech Honors. However, the additional credits and grades thus far earned - by the student shall be included in the grade card but shall not be considered to calculate the CGPA.

- The subjects completed under Honors degree program shall not be considered as equivalent subjects in case the student fails to complete the major degree program.
- Students completed their degree shall not be permitted to register for Honors degree

Structure of Honors in B. Tech. :

- The student shall earn at least 20 credits for award of Honors degree from same branch/department/discipline registered for major degree
- Students can complete Honors degree courses either in the college or online from platforms like NPTEL/SWAYAM etc.
- Of the 20 additional Credits to be acquired, 16 credits shall be earned by undergoing specified courses list in the departments, with four courses, each carrying 4 credits. The remaining 4 credits must be acquired through two NPTEL, which shall be domain specific, each with 2 credits and with a minimum duration of 8/12weeks as recommended by the Board of studies.
- The overall attendance in each semester of regular B. Tech courses and Honors degree courses shall be computed separately
- A student shall maintain an overall attendance of 75% in all registered courses of Honors to be eligible for attending semester end examinations However, condonation for shortage of attendance between 65-75% may be given as per University norms. On the recommendations of College Academic Council, the student concerned will be permitted to take the semester end examinations, Student having less than 65% attendance in Honors courses shall not be permitted for semester end examinations.
- A student detained due to lack of attendance in regular B Tech program shall not be permitted to continue Honors program
- The teaching, examinations (internal and external) and evaluation procedure of Honors degree courses offered in offline is similar to regular B. Tech courses
- Students may choose theory or practical courses to fulfil the minimum credit requirement.
- Students shall be allowed to take maximum two subjects per semester pertaining to their Honors degree
- The students registered for Honors shall not be permitted to register for B. Tech (Minors)

Credits requirement:

- A Student will be eligible to get B. Tech (Honors), if he/she completes an additional 20 credits. These may be acquired either in offline or online like NPTEL/SWAYAM

- The colleges offering Honors degree courses shall be ready teach the courses in offline at their college in the concerned departments. Curriculum and the syllabus of the courses shall be approved by the Board of Studies

- The online NPTEL/SWAYAM subjects selected by a student shall be approved by concerned BOS. The duration of courses shall be a minimum of 12 weeks.

- The assessment and certification of the NPTEL shall be as per the prescribed norms of the NPTEL

- Students shall produce a certificate issued by the NPTEL/SWAYAM conducting agency as a proof of credit attainment.

- The teaching and evaluation procedure of Honors courses offering in offline mode shall be similar to that of regular B. Tech courses

- After successful completion of all major and Honors degree courses with specified CGPA the University will award B. Tech (Honors)

- If a student fails to complete a course offered in online/offline, he/she will not be permitted to continue the Honors degree

Procedure to Applying for Honors degree:

- The department offering the Honors will announce courses required before the start of the session.

- The interested students shall apply for the Honors course to the HOD of the concerned department

- The concerned department will announce the list of the selected students for the Honors.

- The whole process should be completed within one week before the start of every session.

- Selected students shall be permitted to register the courses for Honors degree.

- Each department offering the Honors degree shall submit the final list of selected students to the principal.

- The selected students shall submit a joining letter to the Principal through the concerned HOD.

- The department offering Honors shall maintain the record of student pursing the

Honors degree

➢ With the approval of Principal and suggestion of advisor/mentor, students can choose courses from the approved list and shall register the courses within a week as per the conditions laid down in the structure for the Honor degree.

Allocation of seats for Honors:

➢ The University/institute/colleges will notify the number of the seats for Honor in the concerned department well in advance before the start of the semester

➢ Total number of seats offered for a Honor programme shall be a maximum of 35% of sanctioned intake.

➢ The list of the elective for Honor will be offered from the list of running majors in the concerned subjects. Each department of concerned institute will notify the seats for the minor well before the start of each session as per the following Table

S. No	S. No Name of the course	Sanction seats of major degree programme	Seats offered for minor	Courses offered	Credits for each course

Course Fees for registration of subjects in Major degree :

There is no fee for registration of subjects for major degree program offered in offline at the respective colleges.

Examinations:

(a) The examination for the Honors degree courses offered in offline shall be conducted along with regular B. Tech program.

(b) The examinations (internal and external) and evaluation procedure of Honors degree courses offered in offline is similar to regular B. Tech courses.

(c) A separate transcript shall be issued for the Honor subjects passed in each semester

(d) There is no supplementary examination for the failed subjects in a Honors degree program..

Examination Fees :

Examination Fees will be as per the JNTUK norms.

13. CURRICULAR FRAMEWORK FOR MINOR PROGRAMME:

The goal of introducing B.Tech. Minor is to facilitate the students to choose additional courses from other branches and some advanced subjects of their respective

branch in which they are perusing the degree. This gives a provision to the students to pursue minor other than the discipline in which student got admitted. An aspiring student can choose the courses and laboratories in any other discipline and can get a minor in the chosen specialization in addition to regular major B Tech degree. This way undergraduates are not restricted to learn about courses only in the discipline they get admitted to, but can choose courses of their interest to later on take up a career path of their interest. The students taking up a minor course will get additional credits. A student has to acquire 20 more credits, in addition to 160 credits required for the award of the minor. The department concerned will determine the required courses for award of minor. The subjects in minor program would be a combination of mostly core and some electives of other departments..

OBJECTIVES:

The objectives of initiating the minor are.

- ➢ To diversify the knowledge of the undergraduates
- ➢ To make the undergraduates more employable.
- ➢ To have more educational and professional skills after the completion of his undergraduate courses
- ➢ To give a scope to specialize students in other streams of engineering in addition to the ones they are currently pursuing.

Applicability and Enrolment:

- ➢ To all B. Tech (Regular and Lateral Entry) students admitted in Engineering & Technology
- ➢ The minor will be offered at all J.N.T. University affiliated colleges (Autonomous and Non-Autonomous).
- ➢ There shall be no limit on the number of programs offered under Minor. The minor programs in emerging technologies based on expertise in the respective departments may be offered and minor can also be offered in collaboration with the relevant industries/agencies.
- ➢ Total number of seats offered for a minor program shall be a maximum of 35% sanctioned intake of major degree program.
- ➢ If a minimum enrolments criterion is not met, then the students may be permitted to register for the equivalent MOOC courses as approved by the concerned Head of the department in consultation with BoS.

- The allotment of seat into minor is based on the percentage of marks up to III semester in case of regular entry students and only III semester in case of lateral entry students
- For applicability of minor, both regular B Tech and minor courses shall be successfully completed with specified SGPA/CGPA
- Transfer of credits from a particular minor to regular B. Tech or another major degree and vice-versa shall not be permitted

Entry level:

- The B Tech students (both Regular and Lateral Entry) pursuing a major degree program can register for minor from III semester onwards.
- Students registering for minor shall select the subjects from other branches. For example, if a student pursuing major degree in Electrical & Electronics Engineering shall select the subjects specified for minor in Civil Engineering and he/she will get major degree of Electrical & Electronics Engineering with minor of Civil Engineering.
- Student pursuing major degree in any engineering branch is eligible to register for minor in any other engineering branch. However, students pursuing major degree in a particular Engineering are not allowed to register for minor in the same engineering branch.
- Only those students, who have a CGPA of 8.0 or above, without any backlog, will be permitted to register for a minor
- An SGPA or CGPA in excess of 8.0 has to be maintained in the subsequent semesters in major as well as minor without any backlogs in order to keep the minor registration active.
- Should both the SGPA and CGPA fall below 8.0 at any point after registering for the minor; the minor registration will cease to be active.
- A student registered for minor in a discipline must register and pass in all subjects with a minimum GPA of 8.0 that constitute requirement for award of minor.
- Separate CGPA shall be shown on semester and final transcripts of regular B. Tech and minor.
- Students shall not be permitted to register for minor after completion of VI semester.

- Students shall be permitted to select a maximum of two subjects per semester from the list of subjects specified for minor.
- The students shall complete minor without supplementary appearance within stipulated period as notified by JNTUK for the completion of regular major B. Tech program.
- Minor shall not be awarded at any circumstances without completing the regular major B Tech program in which a student got admitted
- If a student is detained due to lack of attendance, he/she shall not be permitted to register the courses of minor
- If a student failed in any registered course of the minor, he/she shall not be eligible to continue the B.Tech minor. However, the additional, credits and grades thus far earned by the student shall be included in the grade card but shall not be considered to calculate the CGPA
- The subjects completed under minor program shall not be considered as equivalent subjects in case the student fails to complete the major degree program
- Students completed their degree shall not be permitted to register for minor

Structure of Minor in B. Tech:
- The student shall earn at least 20 credits for award of minor from other branch/department/discipline registered for major degree.
- Students can complete minor courses either in the college or in online from platforms like NPTEL/SWAYAM etc.
- The overall attendance in each semester of regular B. Tech courses and minor courses shall be computed separately
- A student shall maintain an overall attendance of 75% in all registered courses of minor to be eligible for attending semester end examinations. However, condonation for shortage of attendance between 65-75% may be given as per University norms. On the recommendations of College Academic Council, the student concerned will be permitted to take the semester end examinations Student having less than 65% attendance in minor courses shall not be permitted for end semester examinations
- A student detained due to lack of attendance in regular B. Tech program shall not be permitted to continue minor program

- The teaching, examinations (internal and external) and evaluation procedure of minor courses offered in offline is similar to regular B. Tech courses
- The students may choose theory or practical courses to fulfill the minimum credit requirement.
- The students may be allowed to take maximum two subjects per semester pertaining to their minor
- The students are permitted to opt for only a single minor course in his/her entire tenure of B.Tech (Engineering)
- The students registered for B. Tech (Hons) shall not be permitted to register for minor
- The student is not permitted to take the electives courses from the parent department to fulfill the minimum credit requirement.

Credits requirement:

- A Student will be eligible to get minor along with major degree engineering, if he/she completes an additional 20 credits. These may be acquired either in offline or online like NPTEL/SWAYAM
- Of the 20 additional credits to be acquired, 16 credits shall be earned by undergoing specified courses of minor, with four courses, each carrying 4 credits. The remaining 4 credits must be acquired through two NPTEL, which shall be domain specific, each with 2 credits and with a minimum duration of 8/12 weeks as recommended by the Board of studies.
- Curriculum and the syllabus of the courses shall be approved by the Board of Studies
- The online NPTEL/SWAYAM subjects selected by student shall be approved by concerned BOS. The duration of courses shall be a minimum of 12 weeks.
- The teaching and evaluation procedure of minor courses offering in offline mode shall be similar to that of regular B. Tech courses
- Students shall produce a certificate issued by the NPTEL/SWAYAM conducting agency as a proof of credit attainment
- The assessment and certification of the NPTEL shall be as per the prescribed norms of the NPTEL.
- After successful completion of all major and minor courses with specified CGPA the University will award both major and minors

> If a student fails to complete a course offered in online/offline, he/she will not be permitted to continue the minor

Procedure to Applying for the Minor :

> The department offering the minor will announce specialization and courses before the start of the session.

> The interested students shall apply through the HOD of his/her parent department

> The concerned department will announce the list of the selected students for the minor.

> The whole process should be completed within one week before the start of every session.

> Selected students shall be permitted to register the courses for minor

Registering for minor courses :

> Each department offering the minor will submit the final list of selected students to the principal.

> The selected students shall submit a joining letter to the Principal through the concerned HOD offering the minor. The student shall inform same to the HOD of his/her parent department.

> Both parent department and department offering minor shall maintain the record of student pursing the minor

> With the approval of Principal and suggestion of advisor, students can choose courses from the approved list and shall register the courses within a week as per the conditions laid down in the structure for the minor.

> If the student wishes to withdraw/change the registration of subject/course, he/she shall inform the same to advisor, subject teacher, HODs of minor department and parent department and Principal within two weeks after registration of the course.

Procedure for Monitoring the Progress of the Scheme:

The students enrolled in the minor courses will be monitored continuously at par with the prevailing practices and examination standards. An advisor/mentor from parent department shall be assigned to a group of students to monitor the progress.

Allocation of seats for minor

- The University/institute/colleges will notify the number of the seats for minor in the concerned department well in advance before the start of the semester
- Total number of seats offered for a minor programme shall be a maximum of 35% of sanctioned intake of major degree programme.
- The list of the elective for minor will be offered from the list of running majors in the concerned subjects. Each department of concerned institute will notify the seats for the minor well before the start of each session as per the following Table

S. No	S. No Name of the course	Sanction seats of major degree programme	Seats offered for minor	Courses offered	Credits for each course

Course Fees for registration of subjects in Minor degree

There is no fee for registration of subjects for minor degree programme offered in offline at the respective colleges.

Examinations

- The examination for the minor courses offered in offline shall be conducted regular along with B. Tech programme.
- The examinations (internal and external) and evaluation procedure of minor courses offered in offline is similar to regular B. Tech courses.
- A separate transcript shall be issued for the minor subjects passed in each semester
- There is no supplementary examination for the failed subjects in a minor program

14. INSTRUCTION DAYS:

A semester shall have a minimum of 90 clear instruction days (including internal examinations).

15. Transfers from other Institutions shall not be permitted.

16. SUPPLEMENTARY EXAMINATIONS:

Supplementary examinations shall be conducted within 4 weeks from the date of announcement of results of regular examinations.

17. WITHHOLDING OF RESULTS: The result of a student shall be withheld

- If the student has not paid the dues, if any, to the institution
- If any case of pending disciplinary action ,
- Involvement in any sort of malpractices etc.
- Involvement in ragging.

18. TRANSITORY REGULATIONS:

a) Detained candidates are eligible for re-admission as and when next offered.

b) The re-admitted candidate will be governed by the rules and regulations under which the candidate has been admitted.

c) In case of transferred students from other Universities, credits shall be transferred to JNTUK as per the academic regulations and course structure of JNTUK.

d) The students seeking transfer to colleges affiliated to JNTUK from various other Universities/ Institutions have to obtain the credits of any equivalent subjects as prescribed by JNTUK. The transferred candidates have to write the backlogs/failed subjects, if any, in the same Institution where he/she was admitted.

19. AMENDMENTS **TO REGULATIONS:**

The Academic Council of MVGR College of Engineering (Autonomous) reserves the right to revise, amend, change or nullify the Regulations, Schemes of Examinations, and/ or Syllabi or any other such matter relating to the requirements of the program which are compatible to the contemporary/emerging trends effectively meeting the needs of society/industry/stake holding groups.

20. <u>Regulations for MALPRACTICES during the conduct of examinations</u>

	Nature of Malpractices/Improper conduct	**Punishment**
1 (a)	If the candidate possesses or keeps accessible in examination hall, any paper, note book, programmable calculators, Cell phones, pager, palm computers or any other form of material concerned with or related to the subject of the examination (theory or practical) in which he is appearing but has not made use of (material shall include any marks on the body of the candidate which can be used as an aid in the subject of the examination)	Expulsion from the examination hall and cancellation of the performance in that subject only. *
(b)	If the candidate gives assistance or guidance or receives it from any other candidate orally or by any other body language methods or communicates through cell phones with any candidate or persons in or outside the exam hall in respect of any matter.	Expulsion from the examination hall and cancellation of the performance in that subject only of all the candidates involved. In case of an outsider, he will be handed

		over to the police and a case is registered against him. *
2	If the candidate has copied in the examination hall from any paper, book, programmable calculators, palm computers or any other form of material relevant to the subject of the examination (theory or practical) in which the candidate is appearing.	Expulsion from the examination hall and cancellation of the performance in that subject and all other subjects the candidate has already appeared including practical examinations and project work and shall not be permitted to appear for the remaining examinations of the subjects of that Semester/year. The Hall Ticket of the candidate is to be cancelled. *
3	If the candidate impersonates any other candidate in connection with the examination.	The candidate who has impersonated shall be expelled from examination hall. The candidate is also debarred and forfeits the seat. The performance of the original candidate, who has been impersonated, shall be cancelled in all the subjects of the examination (including practicals and project work) already appeared and shall not be allowed to appear for examinations of the remaining subjects of that semester/year. The candidate is also debarred for two consecutive semesters from class work and all University examinations. The continuation of the course by the candidate is subject to the academic regulations in connection with forfeiture of seat. If the imposter is an outsider/candidate not on rolls, he will be handed over to the police and a case is registered against him. *
4	If the candidate mishandles the Answer book or additional sheet or takes out or arranges to send out the question paper during the examination or answer book or additional sheet, during or after the examination. Also if the answer script is mutilated / damaged disturbing the shape, of the script, answers, the bar code intentionally.	Expulsion from the examination hall and cancellation of performance in that subject and all the other subjects the candidate has already appeared including practical examinations and project work and shall not be permitted for the remaining examinations of the subjects of that semester. He shall be debarred from class work and all examinations and be allowed to reregistered for the next subsequent odd or even semester only. The continuation of the course by the candidate is subject to the academic regulations in connection with forfeiture of seat.*

5.	Uses objectionable, abusive or offensive language in the answer paper or in letters to the examiners or writes to the examiner requesting him to award pass marks.	The same should be brought to the notice of CE who in turn in consultation with malpractice committee makes decision for cancellation of the performance in that subject.*
6.	Refuses to obey the orders of the Chief Superintendent/Assistant – Superintendent / any officer on duty or misbehaves or creates disturbance of any kind in and around the examination hall or organizes a walk out or instigates others to walk out, or threatens the officer-in charge or any person on duty in or outside the examination hall of any injury to his person or to any of his relations whether by words, either spoken or written or by signs or by visible representation, assaults the officer-in-charge, or any person on duty in or outside the examination hall or any of his relations, or indulges in any other act of misconduct or mischief which result in damage to or destruction of property in the examination hall or any part of the College campus or engages in any other act which in the opinion of the officer on duty amounts to use of unfair means or misconduct or has the tendency to disrupt the orderly conduct of the examination.	In case of students of the college, they shall be expelled from examination halls and cancellation of their performance in that subject and all other subjects the candidate(s) has (have) already appeared and shall not be permitted to appear for the remaining examinations of the subjects of that semester. The candidates also are debarred and forfeit their seats. In case of outsiders, they will be handed over to the police and a police case is registered against them. *
7.	Leaves the exam hall taking away answer script or intentionally tears of the script or any part thereof inside or outside the examination hall.	Expulsion from the examination hall and cancellation of performance in that subject and all the other subjects the candidate has already appeared including practical examinations and project work and shall not be permitted for the remaining examinations of the subjects of that semester/year. The candidate is also debarred for two consecutive semesters from class work and all University examinations. The continuation of the course by the candidate is subject to the academic regulations in connection with forfeiture of seat. *
8.	Possess any lethal weapon or firearm in the examination hall.	Expulsion from the examination hall and cancellation of the performance in that subject and all other subjects the candidate has already appeared including practical examinations and project work and shall not be permitted for the remaining examinations of the subjects of that semester/year. The candidate is also debarred and forfeits the seat. *

9.	If student of the college, who is not a candidate for the particular examination or any person not connected with the college indulges in any malpractice or improper conduct mentioned in clause 6 to 8.	Student of the colleges expulsion from the examination hall and cancellation of the performance in that subject and all other subjects the candidate has already appeared including practical examinations and project work and shall not be permitted for the remaining examinations of the subjects of that semester. The candidate is also debarred and forfeits the seat. Person(s) who do not belong to the College will be handed over to police and, a police case will be registered against them. *
10	Comes in a drunken condition to the examination hall.	Expulsion from the examination hall and cancellation of the performance in that subject and all other subjects the candidate has already appeared including practical examinations and project work and shall not be permitted for the remaining examinations of the subjects of that semester. *
11	Copying detected on the basis of internal evidence, such as, during valuation or during special scrutiny.	Cancellation of the performance in that subject and all other subjects the candidate has appeared including practical examinations and project work of that semester/year examinations.*

*

21. General :
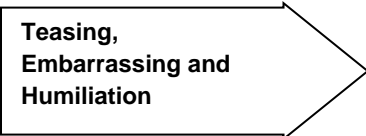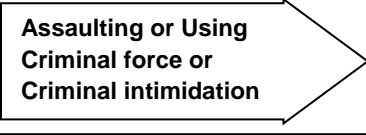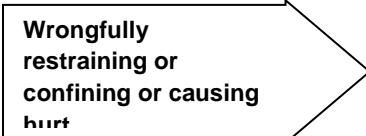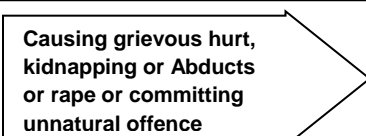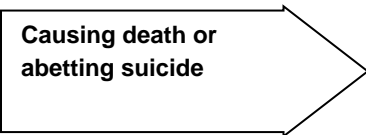
- Wherever the words "he", "him", "his", occur in the regulations, they include "she", "her", "hers".

- The academic regulation should be read as a whole for the purpose of any interpretation.

- In the case of any doubt or ambiguity in the interpretation of the above rules, the decision of the Vice-Chancellor is final.

- The University may change or amend the academic regulations or syllabi at any time and the changes or amendments made shall be applicable to all the students with effect from the dates notified by the University.

✳ ✳ ✳

🚫 Ragging
Prohibition of ragging in
educational institutions Act 26 of 1997

Salient Features

⟹ **Ragging within or outside any educational institution is prohibited.**

⟹ **Ragging means doing an act which causes or is likely to cause Insult or Annoyance of Fear or Apprehension or Threat or Intimidation or outrage of modesty or Injury to a student**

	Imprisonment upto		Fine Upto
Teasing, Embarrassing and Humiliation	**6** Months	+	**Rs. 1,000/-**
Assaulting or Using Criminal force or Criminal intimidation	**1 Year**	+	**Rs. 2,000/-**
Wrongfully restraining or confining or causing hurt	**2 Years**	+	**Rs. 5,000/-**
Causing grievous hurt, kidnapping or Abducts or rape or committing unnatural offence	**5 Years**	+	**Rs. 10,000/-**
Causing death or abetting suicide	**10 Months**	+	**Rs. 50,000/-**

In Case of Emergency CALL TOLL FREE NO. : 1800 - 425 - 1288

LET US MAKE MVGR A RAGGING FREE CAMPUS

🚫 Ragging
ABSOLUTELY NO TO RAGGING

1. Ragging is prohibited as per Act 26 of A.P. Legislative Assembly, 1997.

2. Ragging entails heavy fines and/or imprisonment.

3. Ragging invokes suspension and dismissal from the College.

4. Outsiders are prohibited from entering the College and Hostel without permission.

5. Girl students must be in their hostel rooms by 7.00 p.m.

6. All the students must carry their Identity Cards and show them when demanded

7. The Principal and the Wardens may visit the Hostels and inspect the rooms any time.

PROGRAM STRUCTURE
B. TECH – MECHANICAL ENGINEERING
(A3 Regulation)

SEMESTER-I						
Sl. No	Course Code	Course Title	L	T	P	Credits
1	A3MAT101	Mathematics-I	3	-	-	3
2	A3CYT101	Engineering Chemistry	3	-	-	3
3	A3EET201	Basic Electrical Engineering	3	-	-	3
4	A3CYL101	Engineering Chemistry Laboratory	-	-	3	2
5	A3EEL201	Basic Electrical Engineering Laboratory	-	-	3	2
6	A3MEW201	Workshop	-	-	3	2
7	A3EHA701	Constitution of India	2	-	-	0
Total number of Credits:						**15**

SEMESTER-II						
Sl. No	Course Code	Course Title	L	T	P	Credits
1	A3MAT102	Mathematics-II	3	-	-	3
2	A3PYT101	Engineering Physics	3	-	-	3
3	A3CIT201	Programming for Problem Solving	3	-	-	3
4	A3MED201	Computer Aided Engineering Graphics	1	-	3	3
5	A3PYT101	Engineering Physics Laboratory	-	-	3	2
6	A3CIL201	Programming for Problem Solving Laboratory	-	-	3	2
7	A3EHL001	Essential Communication in English	1	-	3	3
Total number of Credits:						**19**

SEMESTER-III						
Sl. No	Course Code	Course Title	L	T	P	Credits
1	A3EHT001	Effective Technical Communication	2	-	2	3
2	A3EHT002	Professional Ethics and Human Values	3	-	-	3
3	A3CHT101	Biology for Engineers	3	-	-	3
4	A3MAT106	Mathematics-III	3	-	-	3
5	A3MET301	Engineering Mechanics	3	-	-	3
6	A3MET302	Engineering Thermodynamics	3	-	-	3
7	A3MET303	Materials Engineering	3	-	-	3
8	A3MEL301	Computer Aided Geometric Design and Assembly	-	-	4	2
9	A3CHA701	Environmental Science	2	-	-	0
Total number of Credits:						**23**

		SEMESTER-IV				
Sl. No	Course Code	Course Title	L	T	P	Credits
1	A3MAT110	Mathematics-IV	3	-	-	3
2	A3MET202	Design Thinking and Product Innovation	3	-	-	3
3	A3MET304	Strength of Materials	3	-	-	3
4	A3MET305	Fluid Mechanics and Fluid Machines	3	-	-	3
5	A3MET306	Manufacturing Processes	3	-	-	3
6	A3MEI201	AI Tools, Techniques and Applications	3	-	-	3
7	A3MEL302	Materials Lab	-	-	3	2
8	A3MEI201	AI Tools, Techniques and Applications Laboratory	-	-	3	2
9	A3MEP601	Socially Relevant Project	-	-	2	1
10	A3EHA702	Essence of Indian Traditional Knowledge	2	-	-	0
Total number of Credits:						**23**

		SEMESTER-V				
Sl. No	Course Code	Course Title	L	T	P	Credits
1	A3MET201	Industry Internet of Things (IIOT)	2	-	2	3
2	A3MET307	Theory of Machines	3	-	-	3
3	A3MET308	Design of Machine Elements	3	-	-	3
4	A3MET309	Internal Combustion Engines	3	-	-	3
5	A3MET310	Manufacturing Technology	3	-	-	3
6 (PE-1)	A3MET401	Advanced Strength of Materials	3	-	-	3
	A3MET402	Surface Engineering				
	A3MET403	Automobile Engineering				
	A3MET404	Design and Analysis of Experiments				
7 (OE-1)	A3MST002	OE-I: Human Resources Development and Organizational Behavior	3	-	-	3
8	A3MEL303	Thermal Engineering Lab	-	-	3	2
Total number of Credits:						**23**

		SEMESTER-VI				
Sl. No	Course Code	Course Title	L	T	P	Credits
1	A3MET001	Operations Research	3	-	-	3
2	A3MET311	Computer Aided Design and Analysis	3	-	-	3
3	A3MET312	Applied Thermodynamics	3	-	-	3
4	A3MET313	Heat Transfer	3	-	-	3
5 (PE-2)	A3MET405	Design of Transmission Systems	3	-	-	3
	A3MET406	Leadership and Team Management				
	A3MET407	Aircraft and Jet Propulsion				
	A3MET408	Entrepreneurship				
6 (OE-2)	A3XXT5XX		3	-	-	3
7	A3MEL304	Simulation Laboratory	-	-	3	2
8	A3MEP602	Mini Project	-	-	4	2
Total number of Credits:						**22**

		SEMESTER-VII				
Sl. No	Course Code	Course Title	L	T	P	Credits
1	A3MET314	Metrology, Instrumentation and Control Systems	3	-	-	3
2	A3MEI301	Manufacturing Systems	3	-	-	3
3 (PE-3)	A3MET409	Finite Element Analysis	3	-	-	3
	A3MET410	Composite Materials				
	A3MET411	Refrigeration and Air Conditioning				
	A3MET412	Industrial Engineering and Management				
4 (PE-4)	A3MET413	Mechanical Vibrations and Condition Monitoring	3	-	-	3
	A3MET414	Creep, Fatigue and Fracture Mechanics				
	A3MET415	Computational Fluid Dynamics				
	A3MET416	Automation in manufacturing				
5 (PE-5)	A3MET417	Mechatronic Systems	3	-	-	3
	A3MET418	Non Destructive Testing				
	A3MET419	Power Plant Engineering				
	A3MET420	Six Sigma				
6 (PE-6)	A3MET421	Product Lifecycle Management Initiative	3	-	-	3
	A3MET422	Process Planning and Cost Estimation				
	A3MET423	Renewable energy resources				
	A3MET424	Total Quality Management				
2	A3MEI301	Manufacturing Systems Laboratory	-	-	3	1
7	A3MEP603	Project (Phase-I)	-	-	4	2
Total number of Credits:						**21**

Semester - VIII						
S.No	Course Code	Course Title	L	T	P	Credits
1	A3XXT5XX	Open Elective-III (MOOCS)	-	-	-	3
	A3MET507	Waste Heat Recovery and Co-generation				
	A3MET508	Introduction to Nanotechnology				
	A3MET509	Material Characterization Techniques				
2	A3XXT5XX	Open Elective-IV (MOOCS)	-	-	-	3
	A3MET510	Project Management				
	A3MET511	Industrial Safety				
	A3MET512	Advanced Materials				
3	A3MEP604	Project (Phase-II)	-	-	16	8
		Total Number of Credits				**14**

Note: Each department is offering 4 open elective courses. One elective course is from Humanities (Human Resources Development and Organizational Behavior) which is common to all Engineering departments. 2nd Open elective course should be opted from the other departments (List of Open elective courses offered by various departments are given below). 3rd and 4th Open elective courses (Emerging subjects) should be discipline centric.

OPEN ELECTIVES

OPEN ELECTIVE COURSES OFFERED BY THE DEPARTMENT OF CIVIL ENGINEERING						
Sl. No	Course Code	Course Title	L	T	P	Credits
1	A3CET501	Remote Sensing and GIS	3	-	-	3
2	A3CET502	Project Planning and Management	3	-	-	3
3	A3CET503	Road Safety Engineering	3	-	-	3
4	A3CET504	Geomatics	3	-	-	3
5	A3CET505	Building Services	3	-	-	3
6	A3CET506	Water Power Engineering	3	-	-	3
OPEN ELECTIVE COURSES OFFERED BY THE DEPARTMENT OF EEE						
Sl. No	Course Code	Course Title	L	T	P	Credits
1	A3EET501	Basic Control Systems				
2	A3EET502	Applied Electrical Engineering				
3	A3EET503	Electrical Safety				
4	A3EET504	Concepts of Electrical Wiring				
5	A3EET505	Basic Automation Course				
6	A3EET506	Illumination Engineering				

	OPEN ELECTIVE COURSES OFFERED BY THE DEPARTMENT OF MECHANICAL ENGINEERING					
Sl. No	**Course Code**	**Course Title**	**L**	**T**	**P**	**Credits**
1	A3MET501	Introduction to Robotics	3	-	-	3
2	A3MET502	Solar and Wind Energy	3	-	-	3
3	A3MET503	Production and Operations Management	3	-	-	3
4	A3MET504	Micro Electromechanical Systems	3	-	-	3
5	A3MET505	Product Lifecycle Management	3	-	-	3
6	A3MET506	Foundation of Computational Fluid Dynamics	3	-	-	3
	OPEN ELECTIVE COURSES OFFERED BY THE DEPARTMENT OF ECE					
Sl. No	**Course Code**	**Course Title**	**L**	**T**	**P**	**Credits**
1	A3ECT501	Principles of Communication Engineering	3	-	-	3
2	A3ECT502	Electronic Instrumentation	3	-	-	3
3	A3ECT503	Biomedical Engineering	3	-	-	3
4	A3ECT504	Modern Communication Systems	3	-	-	3
5	A3ECT505	Transducers and Sensors	3	-	-	3
6	A3ECT506	Principles of Mobile Communications	3	-	-	3
	OPEN ELECTIVE COURSES OFFERED BY THE DEPARTMENT OF CSE & IT					
Sl. No	**Course Code**	**Course Title**	**L**	**T**	**P**	**Credits**
1	A3CIT501	Fundamentals of Data Structures	3	-	-	3
2	A3CIT502	Object Oriented Programming with JAVA	3	-	-	3
3	A3CIT503	Web Design and Development	3	-	-	3
4	A3CIT504	Python Programming	3	-	-	3
5	A3CIT505	NoSQL Databases	3	-	-	3
6	A3CIT506	Data Analytics	3	-	-	3
	OPEN ELECTIVE COURSES OFFERED BY THE DEPARTMENT OF CHEMICAL ENGINEERING					
Sl. No	**Course Code**	**Course Title**	**L**	**T**	**P**	**Credits**
1	A3CHT501	Computational Fluid Dynamics	3	-	-	3
2	A3CHT502	Non-Conventional Sources of Energy	3	-	-	3
3	A3CHT503	Design & Analysis of Experiments	3	-	-	3
4	A3CHT504	Industrial Waste Water Engineering	3	-	-	3
5	A3CHT505	Green Chemistry & Technology	3	-	-	3
6	A3CHT506	Air Pollution Control and Design of Equipment	3	-	-	3

	SEMESTER - I	L	T	P	C
A3MAT101	**MATHEMATICS-I** **(common to ALL branches)**	**4**	**-**	**-**	**3**
	Total Contact Hours – 48				

SYLLABUS

UNIT-I: LINEAR ALGEBRA-1

Rank of a matrix: Elementary row and column transformations, equivalent matrices, Echelon form of a matrix, calculation of rank by reducing the matrix to Echelon form. System of equations: Linear system of equations, homogeneous and non-homogeneous system of equations, consistency criteria, trivial and non-trivial solutions, solving system of equations by Rank method; Eigenvalues and Eigenvectors: Finding Eigenvalues and Eigenvectors, properties of Eigenvalues and Eigenvectors (statements) including spectral mapping theorem.

UNIT- II: LINEAR ALGEBRA-2

Cayley-Hamilton Theorem: Statement of the theorem and its verification. Applications: Finding higher powers of a matrix, finding matrix polynomials, finding inverse of matrix. Diagonal form of a matrix: Reduction to diagonal form, spectral and modal matrices, finding higher powers of a matrix using diagonalization, Quadratic forms: Matrix form of quadratic forms, orthogonal transformation, canonical form, reduction of quadratic form to canonical form by orthogonal transformation method, rank, index, signature and nature (definiteness) of a quadratic form.

UNIT-III: FIRST ORDER DIFFERENTIAL EQUATIONS & APPLICATIONS

Outlines: Differential Equations(DEs), Order and degree of a DE, Formation of DEs, general solutions of a DE; Solving first order and first degree DEs: linear DEs, Bernoulli's DEs (reducible to linear), exact DEs, integrating factors, non-exact DEs (reducible to exact).

Applications to real world problems: Newton's law of cooling, laws of growth and decay, family of curves, orthogonality of families curves, orthogonal trajectories (Cartesian and polar curves).

UNIT-IV: HIGHER ORDER DIFFERENTIAL EQUATIONS

Differential equations of higher order: Linear differential equations of higher order, its operator form. Solution concepts: General (complete) solution, particular solution. Solution of linear differential equations of higher order: Auxiliary equations, rules for finding complementary functions, rules for finding particular integrals (general and special methods).

UNIT-V: LAPLACE TRANSFORMS

Laplace transformation: Laplace transformation of elementary functions, Properties: Linearity, change of scale, first shifting properties, finding Laplace transformations using properties, Advanced properties: Laplace transformations of derivatives and integrals, multiplication by t^n, division by t (statements), finding Laplace transformations using advanced properties; Inverse Laplace transformations: Finding inverse Laplace transformations using partial fractions, statement of Convolution theorem, finding inverse Laplace transformations by Convolution theorem; Applications: Solving Initial Value Problems by using Laplace transformations.

TEXT BOOKS:

1. B.S. Grewal, Higher Engineering Mathematics, 44/e, Khanna Publishers, 2017
2. T.K.V. Iyengar et al, Engineering Mathematics, S. Chand Publishers, Revised edition

REFERENCE BOOKS:

1. Erwin Kreyszig, Advanced Engineering Mathematics, 10/e, John Wiley & Sons, 2011
2. B.V. Ramana, Higher Engineering Mathematics, Tata McGraw Hill New Delhi, 11th Reprint, 2010
3. T. Veerarajan, Higher Engineering Mathematics, Tata McGraw-Hill, 2008

COURSE OUTCOMES: Learners at the end of this course will be able to

CO 1	Recall the concepts of Linear algebra
CO 2	Recall the solution methods and applicability of first order differential equations
CO 3	Recall the solution methods of higher order differential equations and the concepts of Laplace transforms
CO 4	Use and interpret the concepts of linear algebra
CO 5	Use and interpret solution methods and applicability of first order differential equations
CO 6	Use and interpret solution methods of higher order differential equations and the concepts of Laplace transforms
CO 7	Apply the concepts of linear algebra, differential equations and Laplace transformation to model and solve real world problems

CO/PO Mapping

Course Title:			Mathematics-I (Common to ALL Branches)											
Course Code:			A3MAT101											
Course Designed by			Dept. of Mathematics											
	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1	3	3		2							2			
CO2	3	3		2							2			
CO3	3	3		2							2			
CO4	3	3		2							2			
CO5	3	3		2							2			
CO6	3	3		2							2			
CO7	3	3		2							2			

Course designed by	DEPARTMENT OF MATHEMATICS
Approval	Approved by: Meeting of Board of Studies held on 06.07.2019
	Ratified by: 5th Meeting of Academic Council, 13-07-2019.

	SEMESTER - I	**L**	**T**	**P**	**C**
A3CYT101	**ENGINEERING CHEMISTRY** (Common to all branches)	**4**	**-**	**-**	**3**
	Total Contact Hours – 48				

SYLLABUS

UNIT 1: WATER TECHNOLOGY

Introduction –Soft Water and hardness of water, Estimation of hardness by EDTA Method - Boiler troubles **-** Industrial water treatment – specifications for drinking water, Bureau of Indian Standards(BIS) and World health organization(WHO) standards, zeolite and ion-exchange processes - desalination of brackish water, reverse osmosis (RO) and electro dialysis.

UNIT 2: POLYMERS

Introduction to polymers, functionality of monomers, addition and condensation polymerization, copolymerization, stereospecific polymerization with specific examples. Thermoplastics and Thermo-sets – their differences.

Elastomers – applications with specific examples- Preparation, properties and uses of PVC, Bakelite, Teflon and Nylon-6, 6, Buna-S and Thiokol rubber- Fibre reinforced plastics – carbon fibre, glass fibre and aramids.

UNIT 3: ELECTROCHEMISTRY AND APPLICATIONS

Electrodes – concepts, electrochemical cell, Nernst equation, cell potential calculations.
Primary cells –dry cell- Secondary cells – lead acid, nickel-cadmium and lithium ion batteries- working of the batteries including cell reactions- Fuel cells, hydrogen-oxygen, and methanol fuel cells – working of the cells.
Corrosion: Introduction to corrosion, mechanism of dry and wet corrosion, Pilling Bedworth ratios and uses, Types of corrosion – Differential aeration corrosion, galvanic corrosion, pitting corrosion, waterline corrosion and stress corrosion, Factors affecting the rate of corrosion – metal based factors and environmental based factors, protection techniques – metal coatings – galvanization and tinning, cathodic protection, inhibitors – cathodic and anodic, organic coatings – paints – constituents and their functions.

UNIT-4: CHEMISTRY OF ADVANCED MATERIALS

NANOMATERIALS: introduction- synthesis of Nano material by sol gel method- CVD- engineering applications of Nano materials
CEMENT: Introduction to ordinary Portland cement- manufacturing of OPC- setting and hardening of cement- decay of cement.
FUELS: Introduction- classification- liquid fuels- cracking- knocking- octane number and cetane number; Lubricants- definition- mechanism and properties of lubricants

UNIT 5: INSTRUMENTAL METHODS AND APPLICATIONS

Electromagnetic spectrum. Absorption of radiation: Beer-Lambert's law. Principle, instrumentation (Block diagram and working), applications of UV, IR and NMR spectroscopic methods. Chromatography- introduction- Ion exchange chromatography- applications

Text books:
1. Jain and Jain, Engineering Chemistry, 16/e, Dhanpat Rai, 2013.
2. Skoog and West, Principles of Instrumental Analysis, 6/e, Thomson, 2007.

Reference books:
1. H.F.W. Taylor, Cement Chemistry, 2/e, Thomas Telford Publications, 1997.
2. H.Kaur, Instrumental Methods of chemical analysis, Pragathi Prakashan, 2012.
3. Chemistry for Engineers, Teh Fu Yen, Imperial college press, London

COURSE OUTCOMES:

CO1	The student will have the ability to describe softening methods and desalination processes. He/ She will be able to explain various types of polymers; preparation, properties and engineering applications of thermoplastic, thermosetting plastics, rubbers and FRP's.
CO2	The student will have the ability to describe electrochemical reactions, principles of batteries, fuel cell and corrosion.
CO3	The student will have the ability to outline electromagnetic spectrum and explain the working principles of IR, UV, NMR and chromatographic techniques. The student describes the synthesis, properties and applications of nanomaterials, cement. HE/ She Outlines the cracking methods, knocking of fuels.
CO4	The student will have the ability to differentiate between hard and soft water, demineralization and deionization processes and thermosetting – thermoplastic materials.
CO5	The students will have the ability to give examples on primary and secondary batteries, various types of corrosion, methods of corrosion prevention.
CO6	The student will have the ability to draw inferences on the principles and applications of various instrumental methods and also can compare and contrast between cracking methods.
CO7	The student will have the ability to analyze water samples and validate the results obtained and apply their knowledge on polymers, batteries, materials and instrumentation.

CO/PO Mapping

Course Title:				Engineering Chemistry										
Course Code:				A3CYT101										
Course Designed by				Dept. of Chemistry										
	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1	3								2			1		
CO2	3								2			1		
CO3	3								2			1		
CO4	3								2			1		
CO5	3								2			1		
CO6	3								2			1		
CO7	3								2			1		

Course designed by	DEPARTMENT OF CHEMISTRY
Approval	Approved by: Meeting of Board of Studies held on 29.06.2019
	Ratified by: 5th Meeting of Academic Council, 13-07-2019.

	SEMESTER – I	**L**	**T**	**P**	**C**
A3EET201	**Basic Electrical Engineering (Common to all branches)**	**4**	**-**	**-**	**3**
	Total Contact Hours – 50				

SYLLABUS

UNIT 1: D.C. CIRCUITS
Electrical circuit elements (R, L and C), voltage and current sources, Kirchhoff's current and voltage laws, Analysis of simple circuits with DC excitation, Superposition, Thevenin's and Norton's Theorems, Time-domain analysis of first-order RL and RC circuits.

UNIT 2: A.C. CIRCUITS
Representation of sinusoidal waveforms, Average and RMS values, phasor representation, real power, reactive power, apparent power, power factor, Analysis of single-phase AC circuits (Series & Parallel), Resonance, Three-phase balanced circuits, voltage and current relations in star and delta configurations.

UNIT 3: DC & AC MACHINES [ELEMENTARY TREATMENT ONLY]
Principle and operation of DC Generator - EMF equation – open circuit characteristic of DC shunt generator – principle and operation of DC Motor – Types of DC Motors – Performance Characteristics of DC Motor - Speed control of DC Motor – Principle and operation of single-phase Transformer - OC and SC tests on transformer - principle and operation of single phase & Three phase Induction Motors, construction and working of synchronous motors

UNIT 4:   BASICS OF POWER SYSTEMS:
Layout & operation of Hydro, Thermal, Nuclear Stations - Solar & wind generating stations – Typical AC Power Supply scheme – Elements of Transmission line – Types of Distribution systems: Primary & Secondary distribution systems.

UNIT 5:   ELECTRICAL INSTALLATIONS
Components of LT Switchgear: Switch Fuse Unit (SFU), MCB, ELCB, MCCB, Types of Wires and Cables, Earthing, Types of Batteries, Characteristics of Batteries. Elementary calculations for energy consumption, power factor improvement, battery backup.

TEXT BOOK/ REFERENCES:
1.   D. P. Kothari and I. J. Nagrath, "Basic Electrical Engineering", Tata McGraw Hill, 2010
2.   E. Hughes, "Electrical and Electronics Technology", Pearson, 2010.
3.   Vincent Del Toro, "Electrical Engineering Fundamentals", Pearson, 2015.

COURSE OUTCOMES:

At the end of the course, Student will be able to

CO1	To recall fundamental concepts of electrical circuits such as charge, voltage, current and power.
CO2	Describe the principle of operation of D.C. & A.C. machines.
CO3	Outline the working operation of various generating stations.
CO4	Explain the procedure for solving circuits with A.C and D.C. Excitation
CO5	Summarize the performance characteristics of different machines
CO6	Explain about different equipment used in power industry
CO7	Apply the fundamental laws, associated with Basic Electrical Engineering to solve real world problems in the field of Engineering

CO/PO Mapping

CO / PO mapping	Program Outcomes												PSO1	PSO2
	1	2	3	4	5	6	7	8	9	10	11	12	PSO1	PSO2
CO1	3	3	1	1			3			1			1	1
CO2	3	2	2	2	2					1			1	1
CO3	3	3	3	1	1		1			1			1	1
CO4	3	3	2	1	1		2			1			1	1
CO5	3	3	2	1	1	3	1			1			1	1
CO6	3	3	2	1		2	2			1		1	3	2
CO7	3	3	3	3	3	2	2			2		3	3	3

Course designed by	Department of Electrical & Electronics Engineering
Approval	Approved by: Meeting of Board of Studies held on 29.06.19
	Ratified by: 5th Meeting of Academic Council, 13-07-2019.

	SEMESTER - I	**L**	**T**	**P**	**C**
A3MEW201	**WORKSHOP**	**-**	**-**	**3**	**2**
	Total Contact Hours – 36				

LIST OF EXPERIMENTS

1. Assembly and Disassembly of Bicycle (Fitting)
2. Assembly and Disassembly of Two Wheeler Engine- using power tools (Fitting)
3. Load Estimation for house appliances, Different types of Electric wire specifications, Design of earth pit, Selection of wires and Switch gears.
4. Foundry practice: (Pattern design, Mold making and Casting demonstration)
5. Welding: (Arc welding, Transformer selection and connections, Different "G" position welding. Gas welding)
6. Machine Shop : Turning, Milling, Grinding

TEXT BOOKS:
1. Hajra Choudhury S.K., Hajra Choudhury A.K. and Nirjhar Roy S.K., "Elements of Workshop Technology", Vol. I 2008 and Vol. II 2010, Media promoters and publishers private limited, Mumbai.
2. Kalpakjian S. And Steven S. Schmid, "Manufacturing Engineering and Technology", 4th edition, Pearson Education India Edition, 2002.

REFERENCE BOOKS:
1. Gowri P. Hariharan and A. Suresh Babu,"Manufacturing Technology – I" Pearson Education, 2008.
2. Roy A. Lindberg, "Processes and Materials of Manufacture", 4th edition, Prentice Hall India, 1998.
3. Rao P.N., "Manufacturing Technology", Vol. I and Vol. II, Tata McGrawHill House, 2017.

COURSE OUTCOMES:
After completion of this course, the student will be able to
CO1. Identify and overhaul the components of Bicycle/ Two Wheeler Engine.
CO2. Identify the elements of casting, pattern making and prepare a mould for a single piece and split piece pattern.
CO3. Know the specifications, cutting parameter and perform drilling, milling and grinding operations.
CO4. Know the specifications, welding parameters and perform arc welding and gas welding.
CO5: Calculate load for required electrical design and select correct specifications of electrical requisites.

CO/PO Mapping

Course Title:	Workshop												
Course Code:	A3MEW201												
Course Designed by	Dept. of Mechanical Engineering												
	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1	3				3	1		1	3	2	2	2	3	2
CO2	3		2	1	1	1	1	1	3	2	2	2	3	3
CO3	3		2	1	1	1	1	1	3	2	2	2	3	3
CO4	3		2	1	1	1	1	1	3	2	2	2	3	3
CO5	3	2	2	1	1	3	1	3	3	3	2	2	3	3

Course designed by	DEPARTMENT OF MECHANICAL ENGINEERING
Approval	Approved by: Meeting of Board of Studies held on 29-06-2019
	Ratified by: 5th Meeting of Academic Council, 13-07-2019.

A3CYL101	SEMESTER - I	L	T	P	C
	ENGINEERING CHEMISTRY LABORATORY (Common to all branches)	-	-	3	2
	Total Contact Hours – 36				

SYLLABUS

List of Experiments:
1. Determination of HCl using sodium carbonate
2. Determination of Hardness of a groundwater sample.
3. pH metric titration of strong acid vs. strong base
4. Condcutometric titration of Strong acid VS Strong base
5. Condcutometric titration of Weak acid VS strong base
6. Potentiometric titration of Fe(II) with potassium dichromate
7. Determination of Strength of an acid in Pb-Acid battery
8. Preparation of a polymer
9. Determination of viscosity of polymer solution using survismeter
10. Determination of percentage of Iron in Cement sample by colorimetry
11. Estimation of Calcium oxide in port land Cement
12. Preparation of Nanomaterials (ex: Fe/ Zn/ Ferrite)
13. Adsorption of acetic acid by charcoal
14. Determination of acid value and saponification value of a given lubricant
15. Project based learning (Mandatory for all students)

Course Outcomes:

CO1	The student will be able to determine total hardness, strength of acid in a lead acid battery, calcium in Portland cement using volumetric analysis
CO2	The student will be able to explain condcutometric, potentiometric, pH metric titrations and colorimetric determination
CO3	The student will be able to explain the synthesis of a polymer, nanomaterials

CO/PO Mapping

Course Title:	Engineering Chemistry													
Course Code:	A3CYL101													
Course Designed by	Dept. of Chemistry													
	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1	3		1			1				1	1	2		
CO2	3		1			1				1	1	1		
CO3	3		1							1	1	1		

Course designed by	DEPARTMENT OF CHEMISTRY
Approval	Approved by: Meeting of Board of Studies held on 29.06.2019
	Ratified by: 5th Meeting of Academic Council, 13-07-2019.

A3EEL201	**SEMESTER - I**	**L**	**T**	**P**	**C**
	BASIC ELECTRICAL ENGINEERING LABORATORY (Common to all branches)	**-**	**-**	**3**	**2**
	Total Contact Hours – 36				

SYLLABUS

LIST OF EXPERIMENTS

Basic safety precautions, Introduction and use of measuring instruments – voltmeter, ammeter, multi-meter, oscilloscope, resistors, capacitors and inductors.

1. Verification of Kirchhoff laws.
2. Verification of Network Theorems.
3. Magnetization characteristics of a DC Shunt Generator.
4. Speed control of DC Shunt Motor.
5. Predetermination of performance parameters of 1 – Phase Transformer.
6. I – V Characteristics of Solar PV cell
7. Brake test on DC Shunt Motor.
8. Measurement of earth resistance.
9. Measurement of reactive power in three phase balanced circuit.
10. Measurement of Choke coil parameters
11. Brake test on 3 - Phase Induction Motor.
12. Determination of AC quantities using CRO/DSO.
13. I – V characteristics of battery.

COURSE OUTCOMES:

At the end of the course, Student will be able to

CO1	Identify common electrical equipment used in laboratory.(L1)
CO2	Estimate the ratings of different equipment used to perform an experiment. (L2)
CO3	Demonstrate the usage of various electrical measuring instruments.(L3)
CO4	Analyze the characteristics of rotating & stationery electrical machines (L4).
CO5	Interpret the characteristics of PV cell and Battery.(L5)

CO/PO Mapping

CO / PO Mapping	Program Outcomes													
	1	2	3	4	5	6	7	8	9	10	11	12	PSO 1	PSO 2
CO1	3		1	1	3	1			3	2	2	2	2	1
CO2	3	2	3	3	3	2	1		3	3	2	2	2	3
CO3	2	2	2	2	3	1			3	3	1	2	2	1
CO4	3	3	3	3	2				3	3		2	3	2
CO5	3	3	3	3	3		1		3	3	2	2	3	3

Course designed by	Department of Electrical & Electronics Engineering
Approval	Approved by: Meeting of Board of Studies held on 29.06.19
	Ratified by: 5th Meeting of Academic Council, 13-07-2019.

A3EHA701	SEMESTER - I	L	T	P	C
	CONSTITUTION OF INDIA	2	-	-	0
	Total Contact Hours – 30				

SYLLABUS

UNIT – I: HISTORY OF MAKING OF THE INDIAN CONSTITUTION
History - Drafting Committee,   (Composition & Working)

UNIT – II: PHILOSOPHY OF THE INDIAN CONSTITUTION: Preamble - Salient Features

UNIT-III: CONTOURS OF CONSTITUTIONAL RIGHTS & DUTIES: Fundamental Rights -Right to Equality -Right to Freedom -Right against Exploitation -Right to Freedom of Religion -Cultural and Educational Rights -Right to Constitutional Remedies ; Directive Principles of State Policy ; Fundamental Duties.

UNIT-IV: ORGANS OF GOVERNANCE: Parliament -Composition - Qualifications and Disqualifications - Powers and Functions - Executive - President - Governor - Council of Ministers; Judiciary, Appointment and Transfer of Judges, Qualifications.

UNIT – V: LOCAL ADMINISTRATION: District's Administration head: Role and Importance, Municipalities: Introduction, Mayor and role of Elected Representative, CEO of Municipal Corporation. Panchayati raj: Introduction, PRI: Zila Panchayat. Elected officials and their roles, CEO Zila Panchayat : Position and role. Block level: Organizational Hierarchy (Different departments), Village level: Role of Elected and Appointed officials, Importance of grass root democracy

TEXT BOOK:
Reference Source compilation

REFERENCES:
1. The Constitution of India, 1950 (Bare Act), Government Publication.
2. Dr. S. N. Busi, Dr. B. R. Ambedkar framing of Indian Constitution, 1st Edition, 2015.
3. M. P. Jain, Indian Constitution Law, 7th Edn., Lexis Nexis, 2014.

COURSE OUTCOMES:

CO1.	Students will be able to discuss the growth of the demand for civil rights in India for the bulk of Indians before the arrival of Gandhi in Indian politics.
CO2.	Students will be able discuss the intellectual origins of the framework of argument that informed the conceptualization of social reforms leading to revolution in India.
CO3.	Students will be able to discuss the circumstances surrounding the foundation of the Congress Socialist Party [CSP]under the leadership of Jawaharlal Nehru and the eventual failure of the proposal of direct elections through adult suffrage in the Indian Constitution.
CO4.	Students will be able to discuss the passage of the Hindu Code Bill of 1956.
CO5.	Students will be able to discuss the powers of Executive, Judiciary and Legislature.

CO/PO Mapping

Course Title:	Constitution of India (Common to ALL Branches)													
Course Code:	A3EHA701													
Course Designed by	Dept. of English & Humanities													
	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1						2								
CO2						2								
CO3						2								
CO4						2								
CO5						2								

Course designed by	DEPARTMENT OF ENGLISH & HUMANITIES
Approval	Approved by: Meeting of Board of Studies held on 23.06.15
	Ratified by: 5th Meeting of Academic Council, 13-07-2019.

	SEMESTER - II	**L**	**T**	**P**	**C**
A3MAT102	**MATHEMATICS-II** **(MEC,ECE,EEE,CHE & CIV)**	**4**	**-**	**-**	**3**
	Total Contact Hours – 48				

SYLLABUS

UNIT-I: NUMERICAL METHODS-1

Solving Algebraic and Transcendental Equations: Intermediate value theorem (statement), solution concepts, error in solution, measure of accuracy, approximate and exact solutions, Solution methods: Bisection method, Regula-Falsi method and Newton-Raphson Iterative method; Finite differences: Forward, backward and shift operators, relations among operators, Interpolation: Interpolation and extrapolation, data of equal and unequal intervals, Newton's forward and backward Interpolation formulae, Lagrange's interpolation formula, Fitting polynomials to the data by using Newton's and Lagrange's formulae, Inverse Interpolation by Lagrange's formula.

UNIT- II: NUMERICAL METHODS-2

Numerical Integration: Simpson's and Trapezoidal rules, Weddle's and Boole's rules of integrations; Numerical solutions of ordinary differential equations: Concepts of Initial Value Problem, Taylor's series method, Euler's method, Runge - Kutta method of fourth order; Predictor-corrector method: Milne's method to solve initial value problems.

UNIT-III: MULTIVARIABLE CALCULUS

Overview: Functions of two variables, limit and continuity, partial derivative and its geometrical meaning; Functions of several variables: Partial differential coefficients of higher order, total derivatives, Chain rules for partial differentiation, partial differentiation of Implicit functions; Jacobians: Jacobian and properties, chain rule, functional dependence, Jacobian of implicit functions

Maxima and Minima: Maxima and minima of a function of two variables, constrained maxima and minima, Lagrange's method of undetermined multipliers.

UNIT-IV: PARTIAL DIFFERENTIAL EQUATIONS -FIRST ORDER

Formation of PDEs: Elimination of arbitrary constants, Elimination of arbitrary functions; Solution concepts of PDEs: Complete solution / integral, particular integral, general integral and singular integral, PDEs solvable by direct integration; Linear PDEs of first order (Lagrange's linear equation): Method of grouping and method of multipliers; Nonlinear PDEs of first order: Solution methods of solving PDEs in standard forms I, II, III & IV (as is specified in Text Book 1).

UNIT-V: PARTIAL DIFFERENTIAL EQUATIONS -HIGHER ORDER

Homogeneous Linear Partial Differential Equations of second and higher order with constant coefficients: Symbolic form, Rules for finding complementary function, Rules for finding particular integral, working procedure to get complete solution; Solving nonhomogeneous linear PDEs of second and higher order with constant coefficients; Method of separation of variables: concept of boundary value problem, solving boundary value problems by separating variables.

TEXT BOOKS:
1. B.S. Grewal, Higher Engineering Mathematics, 44/e, Khanna Publishers, 2017
2. T.K.V. Iyengar et al, Engineering Mathematics, S. Chand Publishers, Revised edition

REFERENCE BOOKS:
1. Erwin Kreyszig, Advanced Engineering Mathematics, 10/e, John Wiley & Sons, 2011
2. B.V. Ramana, Higher Engineering Mathematics, Tata McGraw Hill New Delhi, 11th Reprint, 2010
3. T. Veerarajan, Higher Engineering Mathematics, Tata McGraw-Hill, 2008

COURSE OUTCOMES: At the end of course, students shall be able to

CO1	Recall the concepts of numerical methods
CO2	Recall the concepts of multivariable calculus.
CO3	Recall solution methods of PDEs
CO4	Use the concepts of numerical methods to solve equations, do interpolation & numerical integration and also to solve ODEs numerically.
CO5	Use the concepts of multivariable calculus to find maxima & minima of a multivariable function.
CO6	Use solution methods of PDEs to solve BVPs.
CO7	Apply the concepts of numerical methods, multivariable calculus and PDEs to solve real world problems including BVPs.

CO/PO Mapping

Course Title:		MATHEMATICS-II (MEC,ECE,EEE,CHE & CIV)											
Course Code:		A3MAT102											
Course Designed by		Dept. of Mathematics											
	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1	3	3		2							2			
CO2	3	3		2							2			
CO3	3	3		2							2			
CO4	3	3		2							2			
CO5	3	3		2							2			
CO6	3	3		2							2			
CO7	3	3		2							2			

Course designed by	DEPARTMENT OF MATHEMATICS
Approval	Approved by: Meeting of Board of Studies held on 06.07.2019
	Ratified by: 5th Meeting of Academic Council, 13-07-2019.

	SEMESTER – II	L	T	P	C
A3PYT101	**ENGINEERING PHYSICS (COMMON TO CE , ME & CHEM)**	4	-	-	3
	Total Contact Hours – 48				

SYLLABUS

UNIT – I: CRYSTALLOGRAPHY

Introduction- Crystal systems- Bravais lattices- Packing fractions of simple, body centered, face centered cubic structures - Directions and Planes in crystals- Miller indices- Inter planar spacing- Bragg's Law of X-Ray diffraction- Powder X-Ray diffraction method.

UNIT –II: LASER & FIBER OPTICS

LASER: Introduction- Absorption, Spontaneous and stimulated emission of radiation- Einstein coefficients- Population inversion- Basic components of laser- Nd YAG Laser – CO_2 Laser- Applications of LASER.

FIBER OPTICS: Introduction- Principle of optical fiber- Numerical Aperture- Acceptance angle- Classification of optic fibers- Applications of fibers.

UNIT-III: ULTRASONICS & ACOUSTICS

ULTRASONICS-Introduction- Properties of ultrasonic sounds- Generation of Ultrasonic sounds- Magnetostriction- Piezoelectric effect- Detection- Kunts tube- Converse piezoelectric method- Ultrasonic Nondestructive testing technique (pulse-echo technique under reflection mode)- Applications.

ACOUSTICS- Introduction– Reverberation- Reverberation time- Sabines formula for reverberation time- Absorption coefficient and its measurement- Factors effecting acoustic design of hall.

UNIT – IV: THERMODYNAMICS

Introduction- First Law- Isothermal process- Adiabatic process- Work done- Second Law- Carnot's heat engine- Efficiency- Entropy- Physical significance- Entropy and second law- Temperature entropy diagram- Third Law of Thermodynamics- Applications of thermodynamics.

UNIT – V: PRINCIPLES OF MECHANICS

Introduction- System of forces- Resultant of coplanar forces- Method of resolution- Parallel forces- Moment of force- Varignon theorem- Force system in space- Friction- Limiting friction & Impending motion- Coulomb's laws of dry friction- Coefficient of friction- Cone of friction- Types of friction (qualitative).

TEXTBOOKS
1. Engineering Physics by R.K. Gaur and S.L. Gupta, Dhanpat Rai Publications.

REFERENCES
1. RESNICK, HALLIDAY and WALKER, Principles of Physics, Wiley Publishers
2. A.NELSON, Engineering Mechanics: Statics & Dynamics by, Tata Mc Graw Hill Publishers.
3. P.K. NAG, Engineering Thermodynamics, Mc. Graw Hill Publishers

COURSE OUTCOMES:

CO1	The student will be able to recognize the underlying principles of crystalline solids, LASER production and Optical fibers
CO1	The student will be able to gain knowledge on the fundamentals of acoustics and production & detection of ultrasonics
CO3	The student will be able to describe the essentials of thermodynamics, force systems and friction.
CO4	The student will be able to understand crystal structures and X-ray diffraction as a tool for crystal structure analysis.
CO5	The student will be able to understand the importance of industrially relevant LASERS, applications of optical fibers and the prominence of ultrasonics in nondestructive testing.
CO6	The student will be able to understand basic processes involved in thermo-dynamical systems and force systems
C07	The student will have the ability to apply the conceptual knowledge of forces and its related physical quantities in solving engineering problems.

CO/PO MAPPING:

Course Title:		Engineering Physics												
Course Code:		A3PYT101												
Course Designed by		Dept. of Physics												
	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1	3	3						1				1		
CO2	3	3						1				1		
CO3	3	3						1				1		
CO4	3	3						1				1		
CO5	3	3						1				1		
CO6	3	3						1				1		
CO7	3	3						1				1		

Course designed by	DEPARTMENT OF PHYSICS
Approval	Approved by: Meeting of Board of Studies held on 29.06.2019
	Ratified by: 5th Meeting of Academic Council, 13-07-2019.

	SEMESTER – II	**L**	**T**	**P**	**C**
A3CIT201	**PROGRAMMING FOR PROBLEM SOLVING**	**4**	**-**	**-**	**3**
	Total Contact Hours : 54				
	Prerequisites: Mathematics				

SYLLABUS

UNIT – I:
INTRODUCTION: Introduction to Programming, Computer System, Hardware and Software concepts.
PROBLEM SOLVING: Algorithm, Pseudo-code, flow-chart, program development steps, high-level, Assembly and machine languages.
BASICS OF C PROGRAMMING: Structure of C program, identifier, basic data types and sizes, constants, variables, arithmetic operators, relational operators, logical operators, increment and decrement operators, assignment operator, conditional operator, scanf and printf built-in functions, Creating and running programs.

UNIT – II:
BIT-WISE OPERATORS: logical, shift, rotation, masks.
EXPRESSIONS: expressions, type conversions, conditional expressions, precedence and order of evaluation.
SELECTION: Two-way selection: if-else, nested if, examples, multi-way selection: switch, else-if, examples.
ITERATIVE: loops - while, do-while and for statements, break continue, event and counter controlled loops.

UNIT – III:
Part – I:
ARRAYS: Arrays (1-D, 2-D), Character arrays and Strings, Searching (Linear Search and Binary Search).
Part – II: [9 HOURS]
BASIC ALGORITHMS: Basic Sorting Algorithms (Bubble, Insertion and Selection), comparing algorithms for complexity.
FUNCTIONS: Functions, Scope and Extent of Variables, Function Parameters, parameter passing using call-by-value, sub-routines, Storage Classes,#define,#ifdef, #ifndef pre-processor directives.

UNIT – IV:
RECURSION: Definition of Recursion, example programs using recursion like finding Factorial, Fibonacci series, Quick sort, puzzle solving using recursive functions (towers of hanoi, ackerman function).
POINTERS: Definition of Pointers, Pointer Type, Pointer Arithmetic, Function parameter passing using call-by-reference.
MEMORY ALLOCATION: Difference between static and dynamic memory allocation, dynamic memory allocation using built-in functions, dangling pointer, unreferenced memory problem.

UNIT – V:
ENUMERATED, STRUCTURE AND UNION TYPES: Derived types- structures-declaration, definition and initialization of structures, accessing structures, nested structures,

arrays of structures, structures and functions, pointers to structures, self-referential structures, unions, typedef, bit-fields, concept of linked list, program applications.

FILE-HANDLING: Input and output- concept of a file, text files and binary files, Formatted I/O, File I/O operations, command line arguments.

Text Books

1. Programming For Problem Solving, Behrouz A.Forouzan & Richard F.Gilberg, Cengage Publishers, 3rd Edition
2. Programming In C:A Practical Approach, Ajay Mittal, Pearson Education

Reference Books

1. Brian W. Kernighan And Dennis M. Ritchie, The C Programming Language, Prentice Hall Of India
2. Introduction To C Programming, Reema Thareja, Oxford University Press
3. E. Balaguruswamy, Programming In Ansi C, Tata Mcgraw-Hill

COURSE OUTCOMES

The student will

CO1	Have the ability to **describe** a formal algorithmic solution for the given problem, **list** the features of C including scalar & vector data types, operators, **Outline** expressions, expression evaluation, operator precedence, sequential, conditional & iterative constructs.
CO2	Have the ability to **describe** one and two-dimensional arrays, **outline** loops and arrays for searching and **describe** various sorting techniques.
CO3	Have the ability to **outline** the purpose of functions, pointers, command line arguments, dynamic memory allocation. **Define** storage classes. **Describe** command like arguments, structures, unions, and enumeration. Have knowledge of handling files.
CO4	Have the ability to **solve** complex expressions, **design** algorithms and **develop** programs in C language using the basic constructs, data types, operators, control & iterative statements, and arrays.
CO5	Have the ability to **apply** arrays to solve complex matrix related problems and strings. **Compare and contrast** various searching and sorting techniques for complexity.
CO6	Have the ability to **distinguish between** function call types. **Draw inferences on** command line arguments, storage classes, and pre-processor directives. **Use** pointers with functions, arrays, strings, to **solve** complex problems. **Give example** and **solve** classical recursion problems**. Compare and contrast** static and dynamic memory allocation, and **apply** them. **Use** structures and unions to implement and **solve** real-time problems. **Apply** file related functions to process files.
CO7	Have the ability to **Fully appreciate** the art of procedural programming in C and develop programs **optimally** using the full feature set of C language.

Course Title:		Programming for problem solving (Common to ALL Branches)													
Course Code:		A3CIT201													
Course Designed by		Dept. of Computer Science and Engineering													
	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSOM	PSON	PSOO
CO1	3	3						3	2	1		2	1	1	1
CO2	3	3						3	2	1		2	1	1	1
CO3	3	3						3	2	1		2	1	1	1
CO4	3	3	3	3	3	3	3	3	2	1	1	2	3	3	3
CO5	3	3	3	3	3	3	3	3	2	1	1	2	3	3	3
CO6	3	3	3	3	3	3	3	3	3	1	1	2	3	3	3
CO7	3	3	3	3	3	3	3	3	3	1	1	3	3	3	3

Levels of Correlation: High-3, Medium-2, Low-1

	SEMESTER - II	L	T	P	C
A3MED201	COMPUTER AIDED ENGINEERING GRAPHICS	-	-	3	2
	Total Contact Hours – 60				

SYLLABUS

UNIT-I
Overview of Computer Graphics:
Computer technologies that impact on graphical communication, Demonstrating knowledge of CAD software [such as: The Menu System, Toolbars (Standard, Object Properties, Draw, Modify and Dimension), Drawing Area (Background, Crosshairs, Coordinate System), Dialog boxes and windows, Shortcut menus (Button Bars), The Command Line, The Status Bar, Different methods of zoom as used in CAD, Select and erase objects.

Set up of the drawing page and the printer, Scale settings, Setting up of units and drawing limits; ISO and ANSI standards for coordinate dimensioning and tolerancing. Applying dimensions to objects, applying annotations to drawings;

UNIT-II
Layers: Setting up and use of Layers, layers to create drawings, create, edit and use customized layers, concept of view ports.

Introduction to Orthographic Projections: Projections of Points; Projections of Straight Lines parallel to both planes; Projections of Straight Lines-Parallel to one and inclined to other plane.

UNIT-III
Projections of Straight Lines and Planes: Lines inclined to both planes, determination of true lengths, angle of inclinations and traces, Projections of Planes

UNIT-IV
Projections and sections of solids: Projections of simple solids- Sections of solids

UNIT -V
Development of surfaces, Isometric Projection and Conversion of Isometric Views to Orthographic Views: Principles of Isometric projection – Isometric Scale, Isometric Views, Conventions; Isometric Views of lines, Planes, Simple and compound Solids; Conversion of Isometric Views to Orthographic Views and Vice-versa

TEXT BOOKS
1. DM Kulkarni, AP Rastogi, AK Sarkar "Engineering graphics with Auto CAD" PHI Publishers
2. Bhatt N.D., Panchal V.M. & Ingle P.R "Engineering Drawing" Charotar Publishing House.

REFERENCE BOOKS
1. Shah, M.B. & Rana B.C "Engineering Drawing and Computer Graphics", Pearson Education.
2. Agrawal B. & Agrawal C. M "Engineering Graphics", TMH Publication.
3. Narayana, K.L. & P Kannaiah "Engineering Drawing", SciTech Publishers.
4. CAD Software Theory and User Manuals.

COURSE OUTCOMES

At the end of the course the students will be able to:

CO1	Prepare two dimensional drawings using draw and modify commands in Auto CAD software and represent dimensions to the drawings
CO2	Clearly differentiate different types of projections and get solutions to projections of points in Auto CAD by applying the layers concept
CO3	Solve problems related to projections of straight lines and planes
CO4	Prepare simple solids in CAD software and obtain solutions to projections and sections of solids
CO5	Develop the surfaces of simple solids, prepare Isometric drawings and convert isometric drawings into orthographic views

CO/PO Mapping

Course Title:	Computer Aided Engineering Graphics													
Course Code:	A3MED201													
Course Designed by	Dept. of Mechanical Engineering													
	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1	3	2	2		3	1		1	2	3	2	2	3	2
CO2	3	2	2		3	1		1	2	3			2	1
CO3	3	2	2		3	1		1	2	3			2	1
CO4	3	2	3		3	1		1	2	3	2	2	3	1
CO5	3	2	3		3	1		1	2	3	2	2	3	1

Course designed by	DEPARTMENT OF MECHANICAL ENGINEERING
Approval	Approved by: Meeting of Board of Studies held on 29-06-2019
	Ratified by: 5th Meeting of Academic Council, 13-07-2019.

A3PYL101	**SEMESTER – II**	**L**	**T**	**P**	**C**
	ENGINEERING PHYSICS LAB	-	-	3	2
	Total Contact Hours – 42				

LIST OF EXPERIMENTS

1. Determination of size of the micro dimensional system by Laser diffraction.
2. Determination of numerical aperture and acceptance angle of the optic fiber.
3. Determination of lattice constants of the crystal systems.
4. Verification of laws of transverse vibrations in stretched strings by using Sonometer.
5. Determination of velocity of ultrasonic sounds in liquids by acoustic grating method
6. Determination of thermal conductivity coefficient of the disc shaped material.
7. Determination of specific heat of the given liquid by Newton's law of cooling principle.
8. Determination of temperature coefficient resistance for the thermistor.
9. Determination of the static friction coefficient.
10. Determination of rigidity modulus of the wire shaped material by using Torsional pendulum.

TEXTBOOKS:

1. BALASUBRAMANIAN.S, SRINIVASAN.M..N, A Text book of Practical Physics, S Chand Publishers, 2017

REFERENCES:

1. https://vlab.amrita.edu.

COURSE OUTCOMES:

CO1	. Design experiments to determine the size of the micro-dimensional system and the parameters impelling communication through optic fibre.
CO2.	Investigate the powder X-Ray diffraction patterns for crystal structure analysis.
CO3.	Design experiments for demonstration of mechanical resonance and determine the velocity of ultrasonic sounds in liquid media.
CO4.	Design experiments to determine physiognomies of materials like the thermal conductivity coefficient (K), specific heat (s) and temperature coefficient of resistance (α).
CO5	Design experiments to determine the mechanical properties like the rigidity modulus (η) and the static friction coefficient (μ_s).

CO/PO MAPPING:

Course Title:	Engineering Physics Lab												
Course Code:	A3PYL101												
Course Designed by	Dept. of Physics												
	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1 PSO2
CO1	3	3						1	2	1		1	
CO2	3	3						1	2	1		1	
CO3	3	3						1	2	1		1	
CO4	3	3						1	2	1		1	
CO5	3	3						1	2	1		1	

Course designed by	DEPARTMENT OF PHYSICS
Approval	Approved by: Meeting of Board of Studies held on 29.06.2019
	Ratified by: 5th Meeting of Academic Council, 13-07-2019.

	SEMESTER – II	**L**	**T**	**P**	**C**
A3CIL201	**PROGRAMMING FOR PROBLEM SOLVING LABORATORY**	**-**	**-**	**3**	**2**
	Total Contact Hours : 42				
	Prerequisites: Mathematics				

SYLLABUS

WEEK 1:
Objective: Getting familiar with the programming environment on the computer and writing the first program.
Suggested Experiments/Activities:
Tutorial 1: Problem-solving using computers
Lab1: Familiarization with programming environment

    i)   Exposure to Turbo C, gcc, Code Blocks IDE
    ii)   Writing simple programs using printf(), scanf()

WEEK 2:
Objective: Getting familiar with how to formally describe a solution to a problem in a series of finite steps both using textual notation and graphic notation.
Suggested Experiments/Activities:
Tutorial 2: Problem-solving using Algorithms and Flow charts
Lab1: Converting algorithms/flowcharts into C Source code
Developing the algorithms/flowcharts for the following sample programs
    i.   Sum and average of 3 numbers
    ii.   Conversion of Fahrenheit to Celsius and vice versa
    iii.   Simple interest calculation

WEEK 3:
Objective: Learn how to define variables with the desired data-type, initialize them with appropriate values and how arithmetic operators can be used with variables and constants.
Suggested Experiments/Activities:
Tutorial 3: Variable types and type conversions:
Lab 3: Simple computational problems using arithmetic expressions

    i)   Finding the square root of a given number
    ii)   Finding compound interest
    iii)   Area of a triangle using heron's formulae
    iv)   Distance travelled by an object

WEEK 4:
Objective: Explore the full scope of expressions, type-compatibility of variables & constants and operators used in the expression and how operator precedence works.
Suggested Experiments/Activities:
Tutorial 4: Operators and their precedence and associativity:
Lab 4: Simple computational problems using the operator's precedence and associativity
    i)   Evaluate the following expressions
         a. A+B*C+(D*E)+F*G
         b. A/B*C-B+A*D/3
    ii)   a. A+++B---A

b. J=(i++)+(++i)
iii)   Find the maximum of three numbers using conditional operator
iv)   Take marks of 5 subjects in integers, and find the total, average in float

WEEK 5:
Objective: Explore the full scope of different variants of "if construct" namely if-else, null-else, if-else if*-else, switch and nested-if including in what scenario each one of them can be used and how to use them. Explore all relational and logical operators while writing conditionals for "if construct".
Suggested Experiments/Activities:
Tutorial 5: Branching and logical expressions:

Lab 5: Problems involving if-then-else structures

i)   Write a C program to find the max and min of four numbers using if-else
ii)   Write a C program to generate electricity bill
iii)   Find the roots of the quadratic equation
iv)   Write a C program to simulate a calculator using switch case
v)   Write a C program to find the given year is a leap year or not

WEEK 6:
Objective: Explore the full scope of iterative constructs namely while loop, do-while loop and for loop in addition to structured jump constructs like break and continue including when each of these statements is more appropriate to use.
Suggested Experiments/Activities:
Tutorial 6: Loops, while and for loops:
Lab 6: Iterative problems e.g., the sum of series

i)   Find the factorial of given number using any loop
ii)   Find the given number is a prime or not
iii)   Compute sine and cos series
iv)   Checking a number palindrome
v)   Construct a pyramid of numbers

WEEK 7:
Objective: Explore the full scope of Arrays construct namely defining and initializing 1-D and 2-D and more generically n-D arrays and referencing individual array elements from the defined array. Using integer 1-D arrays, explore search solution linear search.

Suggested Experiments/Activities:
Tutorial 7: 1D Arrays: searching
Lab 7: 1D Array manipulation, linear search

i)   Find the min and max of a 1-D integer array
ii)   Perform linear search on1D array
iii)   The reverse of a 1D integer array
iv)   Find 2's complement of the given binary number
v)   Eliminate duplicate elements in an array

WEEK 8:
Objective: Explore the difference between other arrays and character arrays that can be used as Strings by using null character and get comfortable with string by doing experiments that

will reverse a string and concatenate two strings. Explore sorting solution bubble sort using integer arrays.
Suggested Experiments/Activities:
Tutorial 8: 2D arrays, Sorting and Strings
Lab 8: Matrix problems, String operations, Bubble sort

i)   Addition of two matrices
ii)   Multiplication two matrices
iii)   Sort array elements using bubble sort
iv)   Concatenate two strings without built-in functions
v)   Reverse a string using built-in and without built-in string functions

WEEK 9:
Objective: Explore the Functions, sub-routines, scope and extent of variables, doing some experiments by parameter passing using call by value. Basic methods of numerical integration
Suggested Experiments/Activities:
Tutorial 9: Functions, call by value, scope and extent,
Lab 9: Simple functions using call by value, Solving differential equations using Eulers theorem
i)   Write a C function to calculate NCR value
ii)   Write a C function to find the length of a string
iii)   Write a C function to transpose of a matrix
iv)   Write a C function to demonstrate numerical integration of differential equations using Euler's method

WEEK 10:
Objective: Explore how recursive solutions can be programmed by writing recursive functions that can be invoked from the main by programming at-least five distinct problems that have naturally recursive solutions.

Suggested Experiments/Activities:

Tutorial 10: Recursion, the structure of recursive calls
Lab 10: Recursive functions

i)   Write a recursive function to generate Fibonacci series
ii)   Write a recursive function to find the lcm of two numbers
iii)   Write a recursive function to find the factorial of a number
iv)   Write a C Program to implement Ackermann function using recursion
v)   Write a recursive function to find the sum of series.

Objective: Explore the basic difference between normal and pointer variables, Arithmetic operations using pointers and passing variables to functions using pointers
Suggested Experiments/Activities:
Tutorial 11: Call by reference, dangling pointers
Lab 11: Simple functions using Call by reference, Dangling pointers

i)   Write a C program to swap two numbers using call by reference
ii)   Demonstrate Dangling pointer problem using a C program

iii)   Write a C program to copy one string into another using pointer
iv)   Write a C program to find no of lowercase, uppercase, digits and other characters using pointers.

UNIT – V
Objective: Explore pointers to manage a dynamic array of integers, including memory allocation & value initialization, resizing changing and reordering the contents of an array and memory de-allocation using malloc(), calloc(), realloc() and free() functions. Gain experience processing command-line arguments received by C
Suggested Experiments/Activities:
Tutorial 12: Pointers, structures and dynamic memory allocation
Lab 12: Pointers and structures, memory dereference

i)   Write a C program to find the sum of a 1D array using malloc()
ii)   Write a C program to find the total, average of n students using structures
iii)   Enter n students data using calloc() and display failed students list
iv)   Read student name and marks from the command line and display the student details along with the total.
v)   Write a C program to implement realloc()

Objective: Experiment with C Structures, Unions, bit fields and self-referential structures (Singly-linked lists) and nested structures
Suggested Experiments/Activities:
Tutorial 12: Bitfields**,** Self-Referential Structures, Linked lists
Lab 12: Bitfields, linked lists
i)   Read and print a date using dd/mm/yyyy format using bit-fields and differentiate the same without using bit-fields
ii)   Create and display a singly linked list using self-referential structure
iii)   Demonstrate the differences between structures and unions using a C program
iv)   Write a C program to shift/rotate using bitfields
v)   Write a C program to copy one structure variable to another structure of the same type.

Objective: To understand data files and file handling with various file I/O functions. Explore the differences between text and binary files.
Suggested Experiments/Activities:
Tutorial 14: File handling:
Lab 14: File operations

i)   Write a C program to write and read text into a file
ii)   Write a C program to write and read text into a binary file using fread() and fwrite()
iii)   Copy the contents of one file to another file
iv)   Write a C program to merge two files into the third file using command-line arguments
v)   Find no. of lines, words and characters in a file
vi)   Write a C program to print last n characters of a given file.

TEXTBOOKS:
1. Ajay Mittal, Programming in C: A practical approach, Pearson.
2. Byron Gottfried, Schaum's Outline of Programming with C, McGraw-Hill

REFERENCES:
1. Brian W. Kernighan and Dennis M. Ritchie, The C Programming Language, Prentice-Hall of India
2. C Programming, A Problem Solving Approach, Forouzan, Gilberg, Prasad, CENGAGE

COURSE OUTCOMES

CO1.	**Demonstrate** the ability to write a formal algorithmic solution for the given problem, **name & explain** the features of C like types including scalar & vector types, operators, expressions, expression evaluation, operator precedence, sequential, conditional & iterative constructs.
CO2.	**Implement** one and two-dimensional arrays to solve simple mathematical and matrix related problems. **Make use of** loops and arrays for searching and **Compare** various sorting techniques.
CO3.	**Identify** the purpose of functions, pointers, command line arguments, dynamic memory allocation. **Define** storage classes. **Understand** command like arguments, structures and unions. Have **knowledge** of handling files.
CO4.	**Design** algorithms and **develop** programs in C language using the basic constructs, data types, operators, control statements, and arrays.
CO5	**Apply** pointers, functions, derived data types, and dynamic memory allocation, **design** solutions to challenging problems.
CO6	**Illustrate** the art of procedural programming in C and **develop** programs optimally using the full feature set of C language.

Course Title:	Programming for problem solving lab														
Course Code:	A3CIL201														
Course Designed by	Dept. of CSE & IT														
	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO 1	PSO 2	PSO 3
CO1	3	3	1	2	3	1	1	1	2	2			2	2	2
CO2	3	3	1	2	3	1	1	1	2	2			2	2	2
CO3	3	3	2	3	3	2	1	1	2	2			3	3	3
CO4	3	3	2	3	3	3	1	1	2	2			3	3	3
CO5	3	3	3	3	3	3	1	1	2	2			3	3	3
CO6	3	3	3	3	3	3	1	1	3	3	3		3	3	3

Levels of Correlation: High-3, Medium-2, Low-1

Course designed by	DEPARTMENTS OF CSE & IT
Approval	Approved by: Meeting of Board of Studies held on 29-06-2019
	Ratified by: 5th Meeting of Academic Council, 13-07-2019

	SEMESTER - II	**L**	**T**	**P**	**C**
A3EHL001	**Essential Communication in English**	**2**	**-**	**-**	**2**
	Total Contact Hours – 60				

SYLLABUS

UNIT – I: BASIC LANGUAGE SKILLS – A REFRESHER
Organs of Speech: Consonant Sounds &Vowel Sounds; Phonemic Transcription; Using a Dictionary to know the Pronunciation of a word
Presenting Oneself: Introducing oneself -Using different expressions in Formal&Informal Contexts.
Reading a News Article: Identifying the key words and their usage; summarizing the information
Word Study & Mind Mapping: Root words–Derivatives; Homonyms, Homographs, Homophones; Synonyms & Antonyms

UNIT – II: RUDIMENTS OF FUNDAMENTAL COMMUNICATION
The World: Listening & watching Documentaries on World famous Places.
Describing People, Places and Life experiences: Physical Description- Describing someone's qualities – Usage of Jargon to present topography.
Short Story Corner: Reading a short story – Understanding the mood and essence – Sharing different perspectives.
Sentence Patterns: Concord – Rules – Common errors in day-day usage

UNIT-III: COMMUNICATION AT PRACTICE
Oratory Skills: Listening to World's Famous Speeches
JAM (Just a Minute) Talk: Format & Delivery Techniques
Nuances of Language: Company Description –Position Description (Formal) – processes like Chocolate Making(Informal)**.**
Types of Sentences – Declarative, Interrogative, Assertive etc.

UNIT-IV: COMMUNICATION THROUGH CONCEPTUAL LEARNING
BBC English: Watching interviews of Famous people.
Dialogue Practice: Situational Dialogues; Structuring a Role Play
New Inventions: Reading about latest technology pertaining to different fields (Source : Science Journals)
Transformation of sentences: Active Voice-Passive Voice**,** Direct & Indirect Speech**,** Degrees of Comparison**,** Simple Compound & Complex Sentences.

UNIT – V: COMMUNICATION THROUGH LIFE SKILLS
Watching Movies for Language Enrichment & Writing Reviews.
Skits: Enacting a Skit on a Social Issue
Reflections: Reading News Paper Editorial columns, Literacy Reviews, Poetry
Presenting an autobiography: Exploring different styles of writing autobiographies and evolving an own style**.**

TEXT BOOK:
Reference Source Compilation by the Department

REFERENCES:
1. **Fundamentals of Technical Communication** by Meenakshi Raman,OUP.
2. **Living English Structure** by W.Stannard Allen, Pearson Publications.
3. **English Made Easy** by Mary Margaret Hosler, Mc Graw Hill.
4.. **English and Communication Skills for Students of Science and Engineering**, by
   Dhanavel, S.P. Orient Blackswan Ltd.
5. **The Oxford Guide to Writing and Speaking** by John Seely , OUP

COURSE OUTCOMES:

CO1	Student will be able to come to terms with the basic language Skills required to cater to the requirement of the programme undertaken.
CO2	Student will be able to comprehend and analyze the core concepts well.
CO3	Student will be able to gain proficiency in all four skills of Language – Listening, Reading, Speaking and Writing.
CO4	Student will be able to understand the Syntactical and Grammatical Components of English Language and their correct use.
CO5	Student will be able to present his/her ideas confidently in a Professional manner.

CO/PO Mapping

Course Title:	**Essential Communication in English**												
Course Code:	A3EHL001												
Course Designed by	Dept. of English & Humanities												
	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1						2		2	3	3		3		
CO2						2		2	3	3		3		
CO3						2		2	3	3		3		
CO4						2		2	3	3		3		
CO5						2		2	3	3		3		

Course designed by	DEPARTMENT OF ENGLISH & HUMANITIES
Approval	Approved by: Meeting of Board of Studies held on 29.06.19
	Ratified by: 5th Meeting of Academic Council, 13-07-2019.

	SEMESTER - III	L	T	P	C
A3EHT001	**EFFECTIVE TECHNICAL COMMUNICATION**	**2**	**-**	**2**	**3**
	Total Contact Hours – 48				

SYLLABUS

UNIT – I: PROFICIENCY SKILLS IN COMMUNICATION

Listening Comprehension (Basic Level):
- *Working memory – attention –Vocabulary – Inference- comprehension monitoring.*

Elocution:
- *Composition of words in phrases and clauses – Collocation of words – patterns of sentences – proper use of conjunctions.*

Reading Comprehension Practice – I:
- *Reading Passages for Enrichment of Vocabulary and Sentence Improvement.*

Sentence Completion:
- *Concepts &Rules*

UNIT – II: COMMUNICATION FOR COMPETITIVE WORLD

Listening Comprehension- (Advanced):
- *TOEFL – GRE - IELTS Orientation, Mock Tests.*

Group Discussion:
- *Purpose – Planning –Participation. Etiquette – reaching consensus in group work*

Reading Comprehension Practice – II:
- *Skimming & Scanning Techniques*

Idiomatic expressions & Foreign Expressions and their usage

UNIT-III: COMMUNICATION FOR PROFESSIONAL OUTREACH

Interview Skills:
- *Watching Mock Interviews, Interview Training Sessions,*

Mock Interviews :
- *Facing Interviews, Prerequisites and practice*

Cloze Passages :
- *Reading &Understanding the sequence of sentences in passages*

Syllogisms:

- *Major Premise – Minor premise – Conclusion*

Analogies:

- *Types of Analogies*

UNIT-IV: CAREER PLANNING & GUIDANCE

Video Profile:

- *Preparation – Planning - Execution*

Presentation Skills:

- *Making an oral Presentation -Structuring ideas – Power Point Presentation etiquette –Practice*

Reading Comprehension – III (Practice)

- *(Passages culled from model papers of competitive and qualifying examinations)*

Resume Writing & Cover Letter writing

UNIT – V: ENGLISH & PROFESSIONAL ETIQUETTE

Learning through Visuals:

- *Body Language Gestures & Postures.*

Debating Skills:

- *Making an opening statement – rebuttals – Closing statement, Debate etiquette*

Logic based English Language Tests – Practice

Report Writing:

- *Types of Reports – Writing a Technical Report*

TEXT BOOK:

Open Source Compilation

REFERENCES:

1. Basic Communication Skills for Technology by Andrea J.Rutherfoord, Pearson Publications.
2. Business Communication Today Courtland L. Bovee,John V.Thill Abha Chatterjee, Pearson Publications.
3. How to Do Well in GDs and Interviews by Pearson Publications.

COURSE OUTCOMES:

CO1	Student will be able to develop proficiency in Communication in English.
CO2	Student understands the structure and pattern of various competitive and qualifying examinations for higher studies and employment.
CO3	Student will be able to express professionally his/her views to the context.
CO4	Student will be able to understand the need and concept of professional etiquette as a prerequisite for written and spoken communication.
CO5	Student shall be able to hone his/her analytical thinking skills.
CO6	Student will be able to acquire the employability skills needed.

CO/PO Mapping

Course Title:				**Effective Technical Communication**										
Course Code:				A3EHT001										
Course Designed by				Dept. of English & Humanities										
	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1						2		2	3	3		3		
CO2						2		2	3	3		3		
CO3						2		2	3	3		3		
CO4						2		2	3	3		3		
CO5						2		2	3	3		3		
C06						2		2	3	3		3		

Course designed by	DEPARTMENT OF EHGLISH & HUMANITIES
Approval	Approved by: Meeting of Board of Studies held on 29.06.19
	Ratified by: 5th Meeting of Academic Council, 13-07-2019.

A3EHT002	**SEMESTER - III**	**L**	**T**	**P**	**C**
	Professional Ethics and Human Values	2	1	0	3
	Total Contact Hours : 48				
	Prerequisite : UHV				

UNIT I: Need, Basic Guidelines, Content and Process for Value Education

Purpose and motivation for the course, recapitulation from Universal Human Values-I, Self-Exploration, Its content and process; 'Natural Acceptance' and Experiential Validation, Continuous Happiness and Prosperity, Right understanding, Relationship and Physical Facility, Understanding Happiness and Prosperity correctly, Method to fulfil the above human aspirations

UNIT II: Understanding Harmony in the Human Being - Harmony in Myself

Human being as a co-existence of the sentient 'I' and the material 'Body, needs of Self ('I') and 'Body', Body as an instrument of 'I', characteristics and activities of 'I' and harmony in 'I', harmony of I with the Body, Programs to ensure Sanyam and Health.

UNIT III:

Understanding Harmony in the Family and Society- Harmony in Human-Human Relationship

Human-human relationship; meaning of Justice, nine universal values in relationships, Trust and Respect as the foundational values of relationship, Understanding the harmony in the society, Resolution, Prosperity, fearlessness (trust) and co-existence as comprehensive Human Goals, Undivided Society, Universal Order- from family to world family

Harmony in the Nature and Existence

Harmony in the Nature, Interconnectedness and mutual fulfilment among the four orders of nature- recyclability and self-regulation in nature, Understanding Existence as Co-existence, Holistic perception of harmony at all levels of existence

UNIT IV:

Engineering Ethics & Engineers Rights and Responsibilities

The History of Ethics-Purposes for Engineering Ethics-Engineering Ethics- Professional and Professionalism –Professional Roles to be played by an Engineer –Professional Ethics-Types of Inquiry .Safety and Risk, Concept of Safety – Types of Risks - Designing for Safety – Risk- Benefit Analysis-Accidents- Professional Rights and Responsibilities – confidential and proprietary information-Loyalty-Conflict of Interest–Occupational Crimes-industrial espionage-price fixing-endangering lives- Whistle Blowing-types of whistle blowing- Case Studies

UNIT V: Global Issues and IPR

Globalization- Cross-culture Issues- Ethics and Research-Analyzing Ethical Problems in Research-Intellectual Property Rights- Intellectual Property Law- Copyright-principles-

Rights-infringement -Law- Patent – stages-infringement- Law- Case Studies.

*Include practice exercises and case studies will be taken up in practice sessions (tutorial hours)
Ex. To discuss the conduct as an engineer or scientist etc.

Text Books

1. Human Values and Professional Ethics by R R Gaur, R Sangal, G P Bagaria, Excel Books, New Delhi, 2010

Reference Books

1. Jeevan Vidya: Ek Parichaya, A Nagaraj, Jeevan Vidya Prakashan, Amarkantak, 1999.
2. Human Values, A.N. Tripathi, New Age Intl. Publishers, New Delhi, 2004.
3. The Story of Stuff (Book).
4. The Story of My Experiments with Truth - by Mohandas Karamchand Gandhi
5. Small is Beautiful - E. F Schumacher.
6. Slow is Beautiful - Cecile Andrews
7. Economy of Permanence - J C Kumarappa
8. Bharat Mein Angreji Raj - PanditSunderlal
9. Rediscovering India - by Dharampal
10. Hind Swaraj or Indian Home Rule - by Mohandas K. Gandhi
11. India Wins Freedom - Maulana Abdul Kalam Azad
12. Vivekananda - Romain Rolland (English)
13. Gandhi - Romain Rolland (English)

COURSE OUTCOMES	
1.	Knowledge of happiness, prosperity, natural acceptance and harmony
2	Knowledge of harmony in the human being, family, society and nature/existence
3.	Knowledge of professional ethics and global issues & IPR
4.	Understand happiness, prosperity, natural acceptance and harmony
5	Understand harmony in the human being, family, society and nature/existence
6.	Understand the professional ethics, global issues and IPR
7.	Apply universal values and ethics in all spheres of life

A3EHT002 & Professional Ethics And Human Values													
Course designed by													
CO / PO mapping	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
CO1			1			2		3	1				1	
CO2			1			2		3	1				1	
CO3			1			2		3	1				1	
CO4			1			2		3	1				1	
CO5			1			2		3	1				1	
CO6			1			2		3	1				1	
CO7			1	3		2		3	1				1	

A3EHT002 & Professional Ethics And Human Values	
Course designed by	Department of Mechanical Engineering
Approval	Approved by: Meeting of Board of Studies held on 29-06-2019
	Ratified by: 2nd Meeting of Academic Council, 13-07-2019

A3CHT101	**SEMESTER – III**	**L**	**T**	**P**	**C**
	Biology for Engineers	3	0	0	3
	Total Contact Hours – 48				

SYLLABUS

UNIT-I:
Introduction: Biology and its applications, Biological classification, Living Organisms: Cells and Cell theory, Cell structure and function.

UNIT-II:
Biochemistry and molecular analysis: Chemical composition of living forms, analysis of Chemical composition, Carbohydrates, Amino acid and proteins, protein synthesis, Nucleic acids, lipids, nature of bonding and qualitative tests.

Unit-IIIA:
Genetics: Transfer of genetic information, Mendelian Law, Mendel's law of inheritance, Gene interaction, multiple allens, chromosome theory of inheritance., linkage, Recombination, Chromosome mapping, Genetic disorders, Nucleic acids, replication of DNA, types of RNA, Transcription, Genetic code, translation and steps in translation.

Unit-IIIB:
Metabolism: Thermodynamics as applied to biological systems. Exothermic and endothermic reactions. Concept of Keq and its relation to standard free energy, Spontaneity. ATP as an energy currency. The breakdown of glucose to$CO_2 + H_2O$ (Glycolysis and Krebs cycle) and synthesis of glucose from CO_2 and H_2O (Photosynthesis). Energy yielding and energy consuming reactions, Concept of Energy charge.

UNIT-IV:
Enzymes and industrial applications: Mode of action of enzymes, properties of enzymes, chemical reactions, factors affecting enzyme activity, Co-factors, importance of enzymes, industrial application of enzymes.

UNIT-V:
Microbiology and Industrial applications: Microorganism, Growth kinetics, culture media, sterilization, Microscopy, application of microbiology, immunology and immunity, Cancer Biology, stem cells.

Text books:
1. Biology for Engineers by Wiley (ISBN: 9781121439931), 1st edition TMH, New Delhi (2019)
2. Suraish kumar G K, Biology for Engineers, Oxford University Press, New Delhi (2019)

References:

1. Campbell, NA and Reece JB, Biology, International edition, 7th edition or later, Benjamin Cummings, New York (2007 or later)
2. Karp, G, Cell and Molecular Biology: Concepts and Experiments, 7th edition, Wiley, New York (2013)

Course Outcomes:

Students will be able to:

1. Explain the importance of biology in engineering.
2. Identify the importance of chemicals like lipids, sugars, polysaccharides, amino acids and proteins
3. Know the importance of DNA and RNA
4. Describe the process metabolism
5. Know the various applications of industrial enzymes
6. Know the importance of industrial microbiology in the current scenario.
7. Explain importance of the microbes and its applications.

Mapping of POs & COs (Program Outcomes & Course Outcomes)

CO / PO mapping	1	2	3	4	5	6	7	8	9	10	11	12	PSO1	PSO2
A3CHT101 Biology for Engineers														
CO-1	2			2				2	2		2	2	2	
CO-2	2												2	
CO-3	2												2	
CO-4	2												2	
CO-5	2												2	
CO-6	2			2				2	2		2	2	2	
CO-7	2												2	

A3CHT101	Biology for Engineers
Course designed by	Department of Chemical Engineering
Approval	Approved by: Meeting of Board of Studies held on 29th Jun, 2019
	Ratified by: 2nd Meeting of Academic Council, 13th AUG, 2020

	SEMESTER - III	**L**	**T**	**P**	**C**
A3MAT106	**MATHEMATICS-III** **(common to CIV & MEC)**	**3**	**-**	**-**	**3**
	Total Contact Hours – 48				

SYLLABUS

Unit-I: Random Variables & Probability Distributions

Random Variables: Discrete and continuous random variables, properties of mass and density functions. Mathematical Expectation: Properties (statements), Moment Generating Function; Outlines: of Binomial and Poission distributions; Normal Distribution: Probability density function, Normal approximation to Binomial Distribution, Parameters of Normal Distribution(statements), Characteristics of normal distribution, Area under normal curve, Standard normal distribution.

Unit-II: Statistical Methods

Curve fitting by least squares method: Bi-variate data, scatter diagram, method of least squares, normal equations, fitting of straight line, second degree curve (parabola), exponential and power curves; Correlation: types of correlation, measures of correlation, Karl Pearson coefficient of correlation and its properties; Regression Analysis: Regression Coefficients and its Properties, Regression lines.

Unit-III: Multiple Integrals

Double Integral: Concept of double integration, properties, evaluation procedures, change of order of integration, double integrals in polar coordinates; Change of variables: Jacobian of transformations, Change of Cartesian coordinates to polar coordinates in double integrals, Applications of double integrals: Calculation of areas enclosed by plane curves (Cartesian and polar coordinates);

Triple Integrals: Evaluation procedures of triple integrals; Change of variables: Jacobian of transformations, Change of rectangular coordinates to Cylindrical and Spherical polar coordinates in triple integrals; Applications of triple integrals: Volumes of solids.

Unit-IV: Differential Calculus of Vectors

Gradient: Scalar and vector point functions, scalar and vector fields, vector operator 'del', Gradient of a scalar point function (GradΦ), geometrical interpretation of GradΦ, directional derivative, maximum directional derivative, evaluation of scalar potential of an irrotational field; Divergence: Divergence of a vector point function, physical interpretation of divergence, solenoidal vector function; Curl: Curl of a vector point function, physical interpretation of curl, Rotational and Irrotational fields.

Unit-V: Integral Calculus of Vectors

Line integral of a vector function: Line integral and its types, applying line integral to calculate 'circulation' of a fluid particle and total work done by a force; Surface integral of a vector function: Surfaces, types of surfaces, surface integral and its types, evaluation of surface integrals; Volume integral of a vector function: Evaluation of volume integrals; Integral theorems relating line, surface and volume integrals: Green's theorem in a plane, Stoke's theorem and Gauss divergence theorem (all statements), Verification of theorems.

Text Books:

1. B. S. Grewal, Higher Engineering Mathematics, 44/e, Khanna Publishers, 2017
2. T.K.V. Iyengar et al, Engineering Mathematics, S. Chand Publications, Revised edition .

Reference Books:
1. Erwin Kreyszig, Advanced Engineering Mathematics, 10/e, John Wiley & Sons, 2011.
2. B.V. Ramana, Higher Engineering Mathematics, Tata McGraw Hill New Delhi,11th Reprint, 2010.
3. Murugesan and Gurusamy, Probability, Statistics and Random Process, Anuradha Publicatons.

COURSE OUTCOMES

At the end of the course, students shall be able to:

CO 1	Recall the concepts of Random Variables, Probability Distributions, Curve Fitting and Correlation, Regression
CO 2	Recall the concepts of Multiple Integrals
CO 3	Recall the concepts of Vector Calculus
CO 4	Use and Interpret the concepts of Random Variables, Probability Distributions, Curve Fitting and Correlation, Regression
CO 5	Use and interpret the concepts of Multiple Integrals
CO 6	Use and interpret the concepts of Vector Calculus
CO 7	Apply the concepts of Probability Distributions, Statistical Methods, Multiple Integrals and Vector Calculus to model and solve real world problems.

CO/PO Mapping

Course Title:			Mathematics-III (CIV & MEC)											
Course Code:			A3MAT106											
Course Designed by			Dept. of Mathematics											
CO	Program Outcome (PO)													PSO	
	1	2	3	4	5	6	7	8	9	10	11	12	1	2
1	3	3		2							2			
2	3	3		2							2			
3	3	3		2							2			
4	3	3		2							2			
5	3	3		2							2			
6	3	3		2							2			
7	3	3		2							2			

Course Designed by	Dept. of Mathematics
Approval	Approved by Board of Studies (BoS) of Department of Mathematics in its 4th meeting held on 06.07.2019
	Ratified by Academic Council in its 5th meeting held on 13.07.2019.

	SEMESTER - III	L	T	P	C
A3MET301	**ENGINEERING MECHANICS**	3	-	-	3
	Total Contact Hours – 48				

SYLLABUS

UNIT-I Introduction to Engineering Mechanics and System of Forces:
Coplanar force systems: Coplanar forces and its Resultant, Free body diagram, Particle equilibrium and Rigid Body equilibrium; Equilibrium of Coplanar Forces, Moment of Force and its Application, Couples, Types of beams, Static indeterminacy
Non Coplanar/spacial force systems: Components in space, Resultant of spacial force systems, Equilibrium of spacial force Systems in space

UNIT II Basic Structural Analysis & Friction:
Trusses: Types of plane trusses, Perfect plane Trusses, How to determine if a member is in tension or compression; Method of Joints, Method of Sections, Zero force members;
Friction: Types of friction, Limiting friction, Laws of Friction, Impending Motion, wedge friction, Ladder friction.

UNIT III Centroid, Centre of Gravity & Moment of Inertia
Centroid and Centre of Gravity: Centroid of a plane curve, Centroid of a simple plane area from first principle, centroid of composite section, Pappu's theorems, Centre of Gravity and its implications.
Moment of Inertia: Area moment of inertia- Definition, Moment of inertia of plane sections from first principles, Theorems of moment of inertia, Radius of gyration, Moment of inertia of composite sections; Mass moment inertia of simple solids, Mass moment inertia of composite solids

IV Introduction to Kinematics & Kinetics
Introduction to Kinematics: Rectilinear motion, curvilinear motion, Inclined projection, Rotation about fixed axis, Kinematics of plane motion.
Introduction to Kinetics of Rigid Bodies: Basic terms, general principles in dynamics, D'Alembert's principle, kinetics of general plane motion, connected bodies, Kinetics of rigid body rotation.

UNIT V Energy, Momentum Methods & Virtual Work:
Work Energy Method: Work energy principle for translation and its application in plane motion of connected bodies, Work done by a spring, Work energy principle for fixed axis rotation, Work energy method applied to plane motion.
Virtual Work & Momentum Methods: Introductory concepts of Virtual Work, Principle and application of Virtual work. Linear Impulse and momentum, Conservation of momentum, Impact of elastic bodies, Coefficient of restitution.

TEXT BOOKS:
1. S. Timoshenko, D.H. Young, J.V.Rao, Sukumar Pati, Engineering Mechanics, TATA McGraw Hill Education, 2017 .
2. Ferdinand. L. Singer, Engineering Mechanics: Statics & Dynamics, BS Publications, 2011 .

REFERENCE BOOKS:
1. R.C.Hibbeler, Engineering Mechanics - Statics & Dynamics, Pearson, 2016.
2. J.L. Meriam, L.G. Kraige, J.N.Bolton, Engineering Mechanics: Dynamics, Wiley India Edition, 2018.
3. Ferdinand P. Beer, E. Russell Johnston, Vector Mechanics for Engineers, McGraw Hill Publ, 2019.

COURSE OUTCOMES

At the end of the course the student will be able to

i.   Draw free body diagrams for different components of structural/machine members.
ii. Apply the equations of equilibrium for solving the problems of statics.
iii. Determine moment of inertia of any Area/Volume with definable boundaries about any axis.
iv. Differentiate centroid, centre of mass and centre of gravity
v.   Find the displacement, velocity and accelerations of bodies subjected to unbalanced system of forces.
vi. Compare different methods of approach for solving the problems of dynamics
vii. Will be able to apply the knowledge of engineering mechanics for solving the problems of statics and dynamics

A3MET301 ENGINEERING MECHANICS														
Course designed by	Department of Mechanical Engineering													
CO / PO/ PSO mapping	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12	PSO1	PSO2
i	3	3	3	2					1	1	1	2	2	2
ii	3	3	3	2					1	1	1	2	2	2
iii	3	3	3	2					1	1	1	2	1	1
iv	3	3	3	2					1	1	1	2	1	1
v	3	3	3	2					1	1	1	2	2	2
vi	3	3	3	2					1	1	1	2	1	1
vii	3	3	3	2					1	1	1	2	3	3

A3MET301 ENGINEERING MECHANICS	
Course designed by	Department of Mechanical Engineering
Approval	Approved by: Meeting of Board of Studies held on 29-06-2019
	Ratified by: 2nd Meeting of Academic Council, 13-07-2019

	SEMESTER - III	**L**	**T**	**P**	**C**
A3MET302	**ENGINEERING THERMODYNAMICS**	**3**	**0**	**0**	**3**
	Total number of hours (48 lecture hours)				

SYLLABUS

UNIT I
Work and Heat transfer: Fundamentals - System & Control volume; Property, State & Process; Exact & Inexact differentials; Work - Thermodynamic definition of work; examples; Displacement work; Path dependence of displacement work and illustrations for simple processes; electrical, magnetic, gravitational, spring and shaft work. (4)

Temperature, Definition of thermal equilibrium and Zeroth law; Temperature scales; Various Thermometers- Definition of heat; examples of heat/work interaction in systems- First Law for Cyclic & Non-cyclic processes; Concept of total energy E; Demonstration that E is a property; Various modes of energy, Internal energy and Enthalpy. (4)

UNIT II
Properties of Pure substances: Definition of Pure substance, Ideal Gases and real gases, gas mixtures, Compressibility charts- Properties of two-phase systems - Const. temperature and Const. pressure heating of water.

Definitions of saturated states; P-v-T surface; Use of steam tables, Saturation tables; superheated tables; Identification of states & determination of properties, Mollier's chart. (8)

UNIT III
First Law of Thermodynamics: First Law for Flow Processes - Derivation of general energy equation for a control volume; Steady state steady flow processes including throttling; applications of steady flow energy equation.(6)

Second Law of Thermodynamics: Second law - Definitions of direct and reverse heat engines; Definitions of thermal efficiency and COP; Kelvin-Planck and Clausius statements; Definition of reversible process; Internal and external irreversibility; Carnot cycle, Carnot theorem, corollary of Carnot theorem, Absolute temperature scale. (10)

UNIT IV
Entropy: Clausius inequality; Definition of entropy S; Demonstration that entropy S is a property; Evaluation of 'S' for solids, liquids, ideal gases and ideal gas mixtures undergoing various processes; Principle of increase of entropy; Illustration of processes in T-s coordinates, entropy balance to various systems. (8)

UNIT V
Exergy: Availability and Irreversibility, Availability function for systems and Control volumes undergoing different processes, Lost work. second-law efficiency. Exergy balance to closed systems and control volumes. (8)

COURSE OUTCOMES:
After completing this course, the students will be able to
1. Define system, heat, work and pure substance with suitable examples
2. Calculate the heat transfer, work done for different thermodynamic process and Evaluate the thermodynamic properties of steam and gas mixtures
3. State first law and second law of thermodynamics

4. Apply first and second law of thermodynamics to thermal systems
5. Define and demonstrate the concept of entropy and evaluation of s for different fluids
6. Describe the concept of exergy and apply exergy balance for closed and steady flow systems
7. Apply laws of thermodynamics for open and closed systems and calculate properties of pure substances and gas mixtures

TEXT BOOKS:
1. Yunus A. Cengel , Michael A. Boles, *Thermodynamics: An Engineering Approach*, McGraw Hill.2017
2. P.K.Nag, *Engineering Thermodynamics*, McGraw Hill Education. 2017.

REFERENCE BOOKS:
1. Moran, M. J. and Shapiro, H. N., *Fundamentals of Engineering Thermodynamics*, John Wiley and Sons.2014.
2. Sonntag, R. E, Borgnakke, C. and Van Wylen, G. J., *Fundamentals of Thermodynamics*, John Wiley and Sons. 2012.

A3MET302: ENGINEERING THERMODYNAMICS														
Course designed by	Department of Mechanical Engineering													
CO / PO mapping	1	2	3	4	5	6	7	8	9	10	11	12	PSO 1	PSO2
1.	3	3							1	1				
2.	3	3		2	2				1	1				
3.	3	3		2					1	1			1	
4.	3	3		2					1	1			1	
5.	3	3	1	1					1	1		1		
6.	3	2												
7.	3	2			1								2	

A3MET302 : ENGINEERING THERMODYNAMICS	
Course designed by	Department of Mechanical Engineering
Approval	Approved by: Meeting of Board of Studies held on 29-06-2019
	Ratified by: 2nd Meeting of Academic Council, 13-07-2019

A3MET303	**SEMESTER - III**	**L**	**T**	**P**	**C**
	MATERIALS ENGINEERING	3	-	-	3
	Total Contact Hours – 48				

SYLLABUS

UNIT-I

Crystallization: Importance of Material Engineering study, Crystallization &Critical radius. Grain Size Measurement, Grain morphology in casting and welding, Imperfection in solids, Plastic deformation.

UNIT-II

Alloying and Phase diagrams: Need of alloys, Solid solution, Gibb's phase rule, Phase diagrams: Concept of Tie line and Lever rule, Classification of Binary Phase diagrams, Isomorphous Phase diagram, Eutectic Phase diagram, Peritectic, Eutectoid, Strengthening mechanism and Polymorphism, Fe-Fe$_3$C Phase diagram, Problems on Iron-iron carbide phase diagram.

UNIT-III

Ferrous, Non Ferrous Metals and Alloys: Steels and its classification, Steel Designating system, Role of alloying elements in steel, Stainless steel, Tool steel, Hadfield steel, dual phase steel, Maraging steel, Cast iron and its classification, Grey CI, White CI, Malleable and spheroidal CI. Aluminum and its alloys, Copper and its alloys, Titanium and its alloys, Nickel based super alloys

Advanced Materials: Composite materials and their classification with applications, Ceramic materials and classification with applications, Materials for Marine applications, Materials for Aerospace applications, Nano Materials and classification, Bio materials, Concept of DBT & UF grains.

UNIT-4

Bulk Heat treatment: Need for Heat treatment Stages of Heat treatment, Classification, Annealing and types, Normalizing, Construction of TTT, CCT diagram and applications,

Martempring, Austempering, Hardening, Hardenability, Jominy End Quench Test, Tempering, Age hardening, Cryogenic Heat treatment

UNIT-5

Surface Heat treatment: Surface hardening Techniques: Carburizing, Nitriding, Carbo- Nitriding, Cyaniding, Flame hardening, Induction hardening, Plasma hardening, Vacuum Hardening; Surface treatment techniques: Galvanizing, Boronizing, Physical Vapor Deposition, Chemical Vapor Deposition.

TEXT BOOKS:
1. W. D. Callister, Materials Science and Engineering-An Introduction", 6th Edition, Wiley India. 2006
2. Sidney H. Avner: Introduction to Physical Metallurgy, TMH Publishing Co. Ltd. New Delhi,1997.

References:
1. Kenneth G. Budinski and Michael K. Budinski, "Engineering Materials", Prentice Hall of India Private Limited, 4th Indian Reprint, 2002.
2. V.D.Kodgire, Sushil V. Material Science and Metallurgy,43rd Edition, Everest Publishing house,2018.

COURSE OUTCOMES:
Student will be able to
1. Describe crystallization and grain development.
2. Describe phase diagram and its implications for different alloy systems.
3. Select suitable material for an intended application.
4. Classify recently developed materials and describe their applicability in wide areas
5. Select suitable heat treatment process for required mechanical properties of metals
6. Select suitable surface treatment process for specific applications
7. Select suitable material and treatment approach for required structure-property correlation

A3MET303- MATERIALS ENGINEERING														
Course designed by	Department of Mechanical Engineering													
CO / PO mapping	1	2	3	4	5	6	7	8	9	10	11	12	PSO1	PSO2
1.	2	3	-	-	-	-	-	-	2	1	-	1	-	-
2.	2	2	-	-	-	-	-	-	2	1	-	1	-	-
3.	2	3	1	1	-	-	-	-	2	1	-	1	-	-
4.	1	1	1	1	-	-	-	-	2	1	-	1	-	-
5.	3	2	1	1	-	-	-	-	2	1	-	1	-	-
6.	3	2	1	1	-	-	-	-	2	1	-	1	-	-
7.	2	2	1	1	-	1	-	-	2	1	-	1	-	-

A3MET303 MATERIALS ENGINEERING	
Course designed by	Department of Mechanical Engineering
Approval	Approved by: Meeting of Board of Studies held on 29-06-2019
	Ratified by: 2nd Meeting of Academic Council, 13-07-2019

A3MEL301	**SEMESTER - III**	**L**	**T**	**P**	**C**
	Computer Aided Geometric Design and Assembly	-	-	4	2
	Total Contact Hours – 64				

SYLLABUS

Introduction to Modeling packages - ProEngineer, Ideas, CATIA, Uni Graphics, Solid Works.

<u>Drafting</u>
1. Simple 2D-drawing using sketcher options
2. Complex and Application Oriented 2D-drawing using sketcher options

<u>Modeling</u>
3. Simple 3D- drawing using form features
4. Complex 3D-drawing using form features

<u>Assembly</u>
5. Universal Coupling
6. Oldham Coupling
7. Knuckle Joint
8. Cotter joint
9. Eccentric
10. Single plate clutch
11. Square Tool post
12. Clapper block

REFERENCE BOOKS:

1. CATIA V5R14 for Designers by Sham Tickoo
2. Creo Parametric 2.0 by Louis Gary Lamit

COURSE OUTCOMES: At the end of the course the student will be able to

1. Present an overview of CAD and describe its applications in different fields
2. Describe common terms associated with CAD hardware and software.
3. Give outline of basic principles associated with CAD and to demonstrate common drafting and modeling techniques used by professionals.
4. Introduce the advanced capabilities of CAD and how they can be used to increase productivity

CO/PO Mapping

A3MEL301: Computer Aided Geometric Design and Assembly														
Course designed by	Department of Mechanical Engineering														
CO / PO mapping	1	2	3	4	5	6	7	8	9	10	11	12	PSO1	PS02
1.	1		3		3				1	2		2	3	
2.	1		3		3				1	2		2	3	
3.	1		3		3				1	2	1	3	3	
4.	1		3		3				1	2	1	3	3	

A3MEL301: Computer Aided Geometric Design and Assembly	
Course designed by	Department of Mechanical Engineering
Approval	Approved by: Meeting of Board of Studies held on 29-06-2019
	Ratified by: 2nd Meeting of Academic Council, 13-07-2019

A3CHA701	**SEMESTER – III**	**L**	**T**	**P**	**C**
	ENVIRONMENTAL SCIENCE	2	0	0	0
	Total Contact Hours – 30				

COURSE OBJECTIVES	
1.	To study about the scope and importance of multidisciplinary nature of environmental science.
2.	To study about the natural resources and their importance for the sustenance of life and the need to conserve natural resources.
3.	To study about the ecosystem and its function in the environment.
4.	To study about the importance of biodiversity, the threats to biodiversity and conservation practices to protect the biodiversity.
5.	To study about the various types of pollution, its impact and measures to control pollution.
6.	To study about solid waste management techniques
7.	To study about the sustainability nature of environment

ENVIRONMENTAL SCIENCE

SYLLABUS:

UNIT – I:
Multidisciplinary nature of Environmental Studies: Definition, Scope and Importance of Multidisciplinary nature of Environmental Studies, Climate change: Global warming, Acid rains, Ozone layer depletion

UNIT – II:
Natural resources:
Forest resources, deforestation, case studies –Water resources – Use and over utilization of surface and ground water –Floods, drought, conflicts over water, dams – benefits and problems, Mineral resources: Use and exploitation, environmental effects of mining, case studies. Food resources- World food problems, effects of modern agriculture, Land resources-land degradation, soil erosion and desertification, Energy resources: Growing energy needs, renewable and non-renewable energy sources.

UNIT – III:
Part A:
Ecosystem: Concept of an ecosystem**,** Classification, Structure of an Ecosystem: Producers, consumers and decomposers, different functions of an ecosystem.
Part B:
Biodiversity
Definition and types: genetic, species and ecosystem diversity, Values of biodiversity, Conservation of biodiversity: In-situ and Ex-situ conservation of biodiversity.

UNIT – IV:
Environnemental Pollution :
Definition, Cause, effects and control measures of Air pollution, Water pollution, Soil pollution, Marine pollution, Noise pollution, Nuclear hazards. Role of an individual in prevention of pollution.

UNIT – V: Social issues and the environment:
Sustainability, urban and energy related problems
Solid waste Management: Causes, effects and control measures of urban and industrial wastes,

Text Books:

    1. Environmental Studies by Anubha Kaushik, 4th Edition

    2. A Textbook of Environmental Studies by Shaashi Chawla, TMH, New Delhi

    3. Environmental Studies by P.N. Palanisamy, P. Manikandan, A. Geetha, and K. Manjula Rani; Pearson Education, Chennai

References:

    1. Text Book of Environmental Studies by Deeshita Dave & P. Udaya Bhaskar, Cengage Learning.

    2. Environmental Studies by K.V.S.G. Murali Krishna, VGS Publishers, Vijayawada

    3. Environmental Studies by Benny Joseph, Tata McGraw Hill Co, New Delhi

Course Outcomes:

Students will be able to:

    1. Understand the scope and importance of multidisciplinary nature of environmental science.

    2. Understand the natural resources and their importance for the sustenance of life and the need to conserve natural resources.

    3. Understand ecosystem and its function in the environment,

    4. Understand the importance of biodiversity, the threats to biodiversity and conservation practices to protect the biodiversity.

    5. Understand the various types of pollution, its impact and measures to control pollution.

    6. Understand solid waste management technologies.

    7. Understand the sustainability nature of environment.

ENVIRONMENTAL SCIENCE														
CO / PO mapping	1	2	3	4	5	6	7	8	9	10	11	12	PSO1	PSO2
CO-1	2					1								
CO-2	1					2	2							
CO-3	2					2	1						1	
CO-4	1					1								
CO-5	1	1					1						1	
CO-6	1					2	1						1	
CO-7	1	1											1	

A3CHA701	ENVIRONMENTAL SCIENCE
Course designed by	Department of Chemical Engineering
Approval	Approved by: Meeting of Board of Studies held on 29th Jun, 2019
	Ratified by: 2nd Meeting of Academic Council, 13th AUG, 2020